

An Atlas of the Infra-Red Solar Spectrum from 1 to 6\$\cdot \$5 \$\mu \$ Observed from a High-Altitude Aircraft

J. T. Houghton, N. D. P. Hughes, T. S. Moss and J. S. Seeley

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AN ATLAS OF THE INFRA-RED SOLAR SPECTRUM FROM 1 TO 6.5μ OBSERVED FROM A HIGH-ALTITUDE AIRCRAFT

By J. T. HOUGHTON,* N. D. P. HUGHES, T. S. MOSS AND J. S. SEELEY[†]

Royal Aircraft Establishment, Farnborough, Hants

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Records are presented of the infra-red solar spectrum from 1 to 6.5μ , observed from altitudes up to 15 km. A resolution of about 1 cm^{-1} has been obtained over the whole region and 1200 absorption lines belonging to water vapour, CO₂, CO, N₂O and CH₄ have been identified.

1. INTRODUCTION

The infra-red solar spectrum has been the subject of intensive study from observatories on the earth's surface (see, for example, Migeotte, Neven & Swensson 1956). Large portions of the infra-red region, however, are completely obscured, even at high mountain observatories, by absorption bands of water vapour and carbon dioxide. At altitudes above 40000 ft., which are easily reached by modern jet aircraft, the water vapour absorption bands are largely transparent and the carbon dioxide absorption bands are much reduced, so that more of the solar spectrum is available for study. This project was initiated by Dr F. E. Jones in consultation with the Gassiot Committee of the Royal Society as a contribution to the programme of the International Geophysical Year.

This atlas of the infra-red solar spectrum from 1 to 6.5μ is a record of observations made during ninety-eight flights with a high-resolution grating spectrometer installed in a Canberra aircraft of the Royal Aircraft Establishment, Farnborough, at altitudes of 20000 to 49000 ft. The instrumentation, which has been described in detail by Houghton, Moss & Chamberlain (1958), was, briefly, as follows. A gauze-covered hole in the unpressurized part of the aircraft fuselage allowed sunlight to fall on a plane mirror 6 in. square, which formed part of a sun-following system. The spectrometer, which had an f/6 optical system, used a prism pre-monochromator and a 7500 lines/in. diffraction grating, 4 in. \times 3 in. in size. A resolution of about 1 cm⁻¹ was obtained within the wavelength range 1 to 6.5μ . Lead salt photoconductive detectors were used for the following spectral ranges: 1 to 3μ , uncooled lead sulphide; 3 to 5μ , cooled lead telluride, 5 to 6.5μ , cooled lead selenide. A few results were also obtained with a gold-doped germanium detector. Great care was taken to ensure that the spectrometer was free of water vapour or other absorbing gases.

The spectrum was scanned by rotating the grating while the signal from the detector amplifier was recorded by a multi-channel film recording galvanometer. The galvanometer recorded on photographic paper, 60 mm wide, moving usually at $\frac{1}{4}$ in./s, but at $\frac{1}{16}$ in./s for the longest wavelengths. During preliminary work a potentiometer penrecorder was used, but the spectra presented here were all made with the film recorder,

> * Now at Clarendon Laboratory, Oxford. † Now at Queen Mary College, London.

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which had the advantage of a much faster response. Additional channels indicated the output from a photocell monitoring the total sunlight entering the spectrometer and the angular position of the grating at intervals of 3 minutes of arc. The latter marks, which enabled a fairly accurate wavelength scale to be established, have been removed from the records which are reproduced in the atlas at approximately the original size.

Variations in the monitoring signal were due either to imperfect functioning of the sun-follower in compensating for aircraft movements or to variations in the degree of obscuration of the gauze 'window'. The variations in the monitor and detector signals are not necessarily in proportion and in general the monitor signal shows greater variation than the spectral record, probably because the former is determined mainly by the peak solar radiation which, for example, is much more susceptible to scattering by traces of high cloud than the infra-red radiation. Where the monitor shows little variation the spectra can be used directly to determine absolute absorption levels; where variations occur, careful comparison of the two traces will still enable a background energy curve to be drawn over small wavelength intervals in many cases.

2. The figures

The region from 1 to 6.5μ has been divided into 35 sections, with a small overlap on each section. Wherever possible, an opening shows four observations of the same section, together with the appropriate portion of the table of line identification (table 2).

In general the three upper traces, on the left-hand page, show records obtained from the aircraft, divided into the altitude categories: (a) above 40000 ft., (b) 30000 to 40000 ft. and (c) 20000 to 30000 ft. On the right-hand page a ground level record (d), obtained near Farnborough (240 ft. above sea level) with the same instrument has been included for comparison. In the spectral regions where no solar radiation reaches the ground, suitable laboratory spectra are included in place of the solar spectrum (d). Table 1, pp. 51 and 52, identifies the conditions under which the spectra in each figure were observed.

Each section comprises about 8 in. of the original record, which for figures 1 to 33 would have taken 30 s to scan. For figures 34 and 35, which were recorded at the slower paper speed, the time would have been 120 s.

The monitor signal is usually the broader line on the trace. The zero level of the detector galvanometer is shown by the base line which has been drawn in, and a short example of the noise level encountered when the detector was not illuminated has been included at the end of the trace where appropriate. The spectral bandwith used in each observation is represented by the gap in the symbol $\neg \vdash$. A wave-number scale has been constructed for each page; small fluctuations in the recording speed often occurred and since these cannot be corrected the scale is to be used only as a guide.

3. The identifications

The most prominent features in each section have been marked with a vertical dash on one, or sometimes two, of the traces in each figure. Every tenth dash is numbered to correspond to table 2. In identifying these features the frequency was first established to

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within about 5 cm⁻¹ by means of the grating rotation markers. Careful comparison of the records with data from other sources then enabled the identifications in table 2 to be listed. The wave numbers (corrected to vacuum) given in this table are quoted from the various sources noted by a number in the reference column. These numbered references have been listed on p. 53. Particular use has been made of the three photometric atlases covering the region studied in the present work: from 1 to 2.5μ , Mohler, Pierce, McMath & Goldberg (1950), at the Mount Wilson Observatory; 2.8 to 5.3μ , Migeotte *et al.* (1956) at the Jungfraujoch Observatory; and Shaw, Chapman, Howard & Oxholm (1951) at Columbus, Ohio.

Most of the spectra in this atlas are not so well resolved as those given in the sources quoted; many of the features, registered with a single mark, are blends of lines quite clearly resolved by other workers. In each of these cases the wave number assigned to the feature is that of the strongest component, which is listed first in the table. Where the intensities are approximately equal the identifications are given in the order of the wave numbers. The ν_2 band of water vapour near 6μ , however, is revealed in greater detail in the high-altitude solar spectra than in published laboratory spectra. The additional features have been given an interpolated wave number from the known positions of nearby lines.

4. DISCUSSION

The prominent features of the spectra are already well known (see, for example, Goldberg 1954), being bands due to the atmospheric constituents: water vapour, carbon dioxide, carbon monoxide, methane and nitrous oxide. Quantitative studies of some of the water-vapour lines have been made by Houghton & Seeley (1960), and of some of the lines of methane, nitrous oxide and carbon monoxide by Seeley & Houghton (1961).

Some lines due to absorption in the solar atmosphere are prominent in the highaltitude observations and are marked with the symbol \odot in table 2. Among these are three absorption lines of hydrogen which are obscured by water vapour in the groundlevel spectrum and have not previously been observed. They are the second (4-6 transition) and fourth (4-8) members of the Brackett series and the first member (3-4 transition) of the Paschen-Ritz series. The line 4-6 occurs at 3808 cm⁻¹ and overlaps the water-vapour line at 3806.7 cm⁻¹ (figure 11, no. 24); it can be distinguished by comparing the solar spectra with figure 11 (d), which is a record of the residual water-vapour absorption in the spectrometer at 45000 ft. obtained by mounting a small tungsten filament lamp at the entrance slit. The other two new hydrogen lines appear on figure 6, lines no. 26 and no. 52.

The *R*-branch of the v_3 band of methane, previously obscured by absorption due to the $2v_2$ water-vapour band, is revealed clearly in the high-altitude observations in figure 16. In figure 22 the v_3 band of carbon dioxide is the only extended region of complete absorption which remained at the highest altitude (49000 ft.). The short-wavelength absorption edge is advanced by about 5 cm⁻¹ towards the band centre at this altitude.

There are a number of spectral features listed in table 2 for figures 23 and 24 which have a separation *less* than the spectral bandwidth of the spectrometer. Calculations show that the bandwidth used is sufficient to give the partial resolution observed. The repetition of these features in the several observations confirms their reality.

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Laboratory spectra, obtained with a short path (6 ft. of air at atmospheric pressure) are included in figures 31 to 35, since they have been helpful in identifying the new features observed at altitude in this region.

The data in the atlas are the culmination of a comprehensive flying programme, during which many members of the engineering services and aircrew associated with R.A.E. Radio Flight have given us their patient co-operation. We are grateful to our colleagues Mr I. D. Birch, who kept the equipment in working order and Mr T. D. F. Hawkins who acted as observer on many flights, and for assistance by our printing department in preparing the figures.

The lead selenide detector with a bloomed silicon window and the gold-doped germanium detector were generously provided by Dr B. Bode, Santa Barbara Research Centre, and Dr H. Levinstein, Syracuse University, respectively.

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TABLE I

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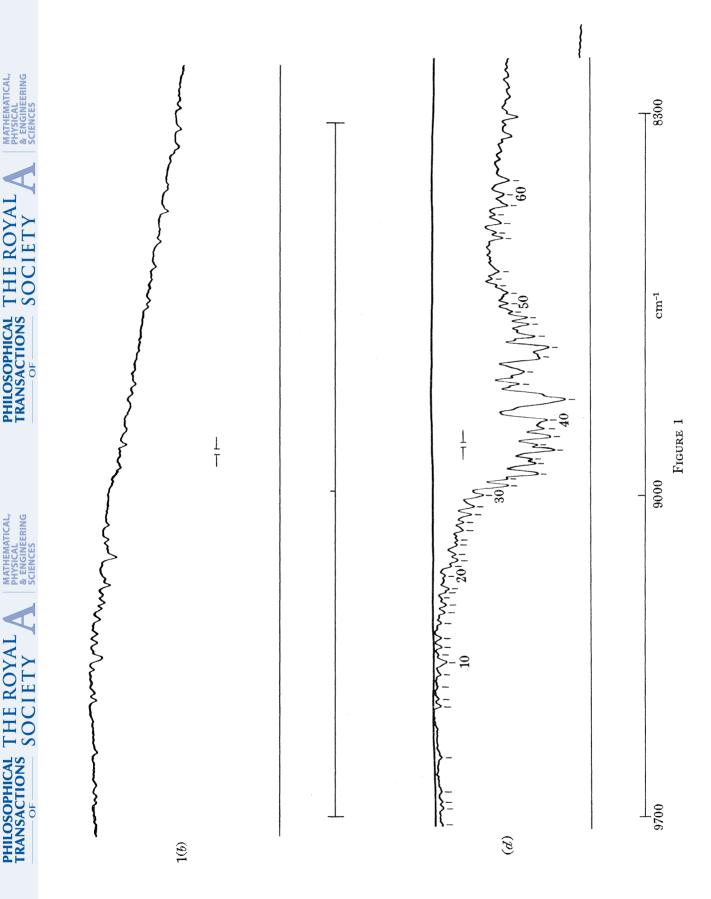


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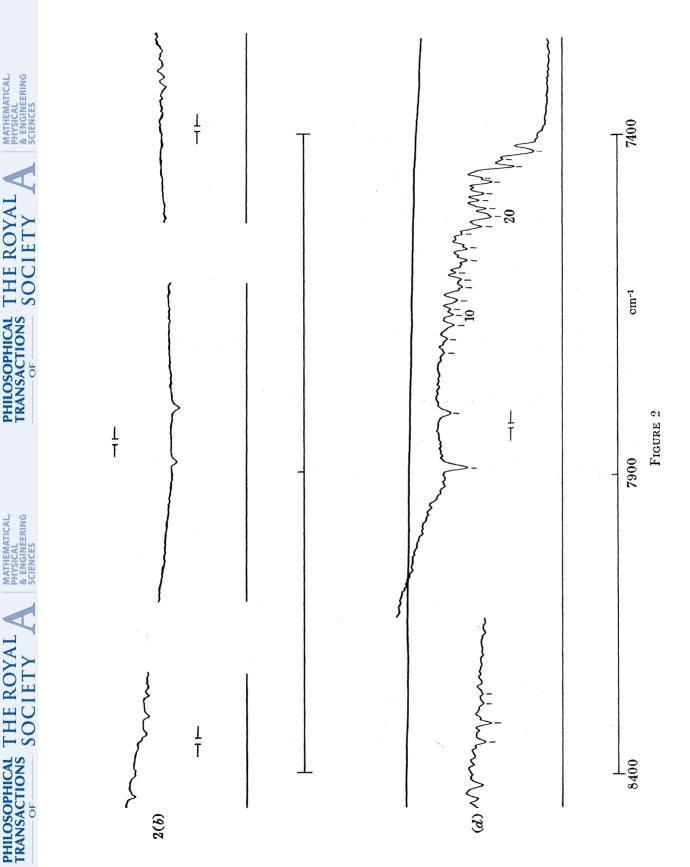
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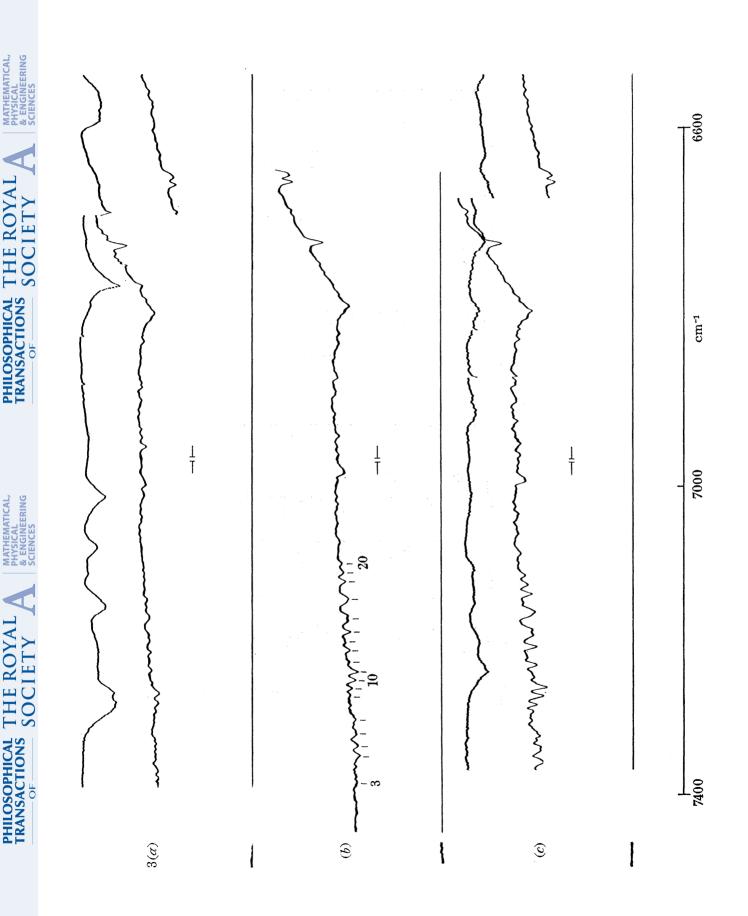


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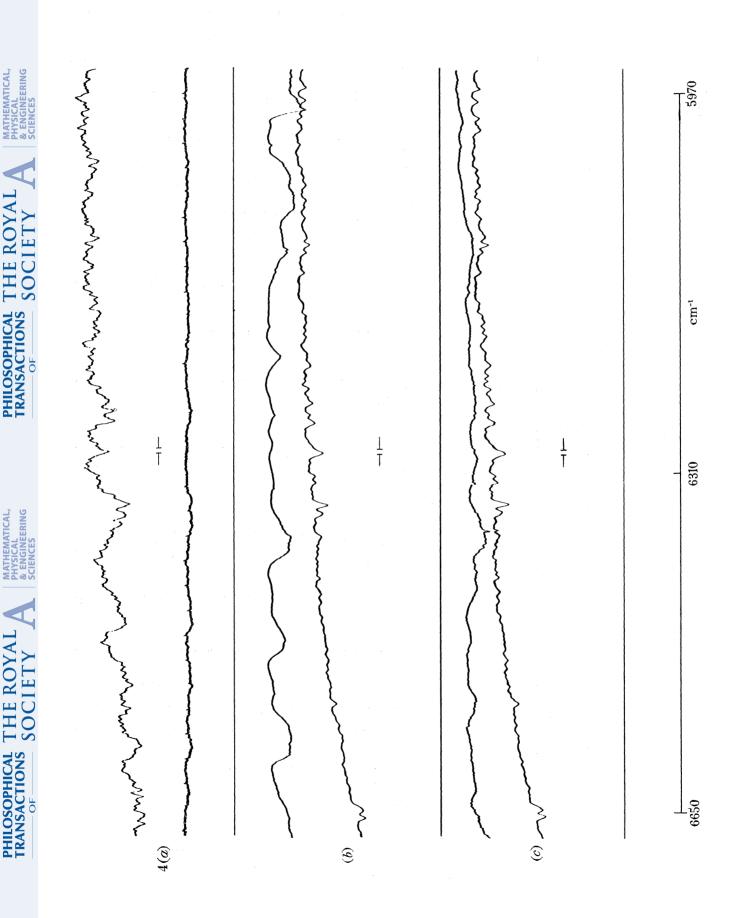
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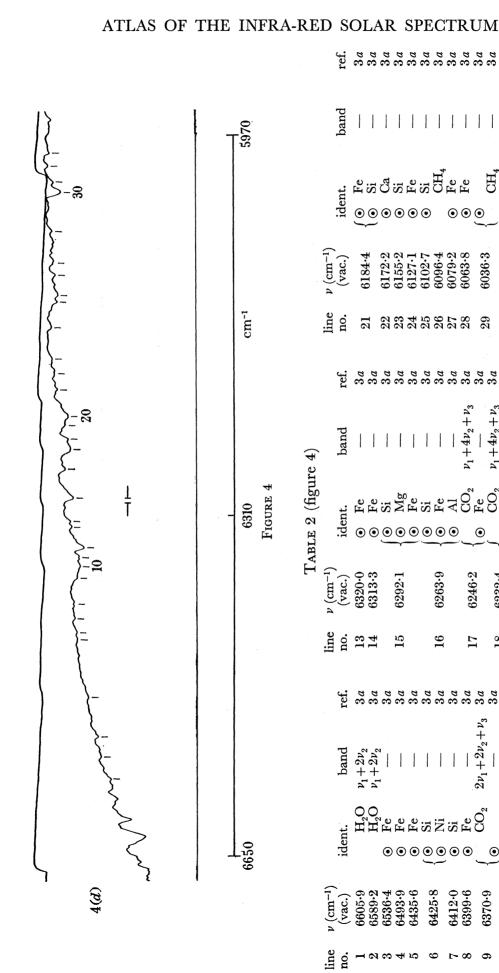
ēf. $\nu_1+\nu_3,\,2\nu_3$ $\begin{array}{c} 2\nu_3\\ 2\nu_3\\ \nu_1+\nu_3, 2\nu_3\\ \nu_1+\nu_3, 2\nu_3\\ \nu_1+\nu_3, 2\nu_3\\ \nu_1+\nu_3, 2\nu_3\\ \nu_1+\nu_3, 2\nu_3\\ \nu_1+\nu_3, 2\nu_3\\ 2\nu_1, \nu_1+\nu_3\\ 2\nu_2, \nu_2\\ 2\nu_1, \nu_1+\nu_3\\ 2\nu_2, \nu_3\\ 2\nu_1, \nu_1+\nu_3\\ 2\nu_2, \nu_1+\nu_3\\ 2\nu_1, \nu_1+\nu_2\\ 2\nu_1, \nu_1+\nu_2\\ 2\nu_1, \nu_1+\nu_3\\ 2\nu_1, \nu_1+\nu_2\\ 2\nu_2, \nu_1+\nu_2\\ 2\nu_1, \nu_1+\nu_2\\ 2\nu_2, \nu_1+\nu_2\\ 2\nu_1, \nu_1+\nu_2\\ 2\nu_2, \nu_1+\nu_2\\ 2\nu_1, \nu_1+$ band ident. ٩ $\nu \stackrel{\nu (cm^{-1})}{(vac.)}$ 7593.0 7575.1 7557.2 7557.2 7556.0 7536.0 7536.0 75311.0 7511.0 7495.4 7495.4 7484.9 7486.4 7486.4 7461.2 7481.2 7481.2 7481.2 7481.2 7481.2 7481.2 7481.2 7481.2 7481.2 7481.2 7481.2 7601.9 TABLE 2 (figure 2) line no. 14 ref. band Санана Синина Синина Синина Состо Санана Состо Санана Санана Состо Санана Санана Состо Санана Состо Сос Сос Сос Состо Сос Сос Сос Состо С ident. $\underline{\circ}$ ۲ \odot 0 ۲ $\nu \, (\rm cm^{-1})$ 8255.2 7881.3 7799.3 7704.2 7665.9 7667.0 7653.1 7653.1 7633.8 8341.7 8307.4 8273-4 7612-0 (vac.) line no. -3 က 4 2 2 4 7 6 2 4 111 0 9 8 7 6 7 4 13 111 0 9 8 7 6 7 4



ATLAS O	F THE	INFF	RA-	RE	D	SO]	LA]	R	SPI	ECJ	[R	UM	1			ł	59
			ref.	3a 3a	3 <i>a</i> 9	3 9 6 3 7 6	3a 3a	3a	оа 3а	3a 3a	3a	3a 3a	3a 3a	30 a 30 a 30 a	3 <i>a</i>	3a 3a	3a 3a
			band	$2\nu_2 + \nu_3 \\ 2\nu_5 + \nu_3$	$2\nu_2 + \nu_3$	$2v_2 + v_3$ $2v_2 + v_3$	$2\nu_{2} + \nu_{3}$ $2\nu_{5} + \nu_{3}$		$2\nu_2 + \nu_3 \\ 2\nu_2 + \nu_3$	$\nu_1 + 2\nu_2$	$\nu_1 + 2\nu_2$	$2\nu_2 + \nu_3$	$ \nu_1 + 2\nu_2 $	$\nu_1 + 2\nu_2$	$2\nu_2 + \nu_3$	$v_1 + 2v_2 + 2v_2 + 2v_2$	
MATHEMATICAL, PHYSICAL SCIENCES SCIENCES	6600		ident.	Н <u>,</u> 0 Н,0	H ₂ O	H ² O		(⊙ Mg	H2O	${\rm (O)}_{\rm Fe}^{\rm H_2O}$		€ H ₂ O Fe	$\int H_2 O$			H ² OO	H20 H20
IETY ROYAL A		$\nu (\mathrm{cm}^{-1})$	(vac.)	6775.9 6766.5	6755-0	6736.8	1-92/9	4.0Z/0	6112.1	6685.3	6675-0	6658-3	6653-7	6647.0	$6634 \cdot 5$		6605•9 6589•2
		line	no.	48 49	50	22	53	0 4 л	56 56	57	58	59	09	61	62	63 64 7	65 66
	cm ⁻¹		ref.	3a 3a	3a	3 a 3 a	3a 3a	3 <i>a</i>	3a 3a	3a 3a	3a	3a 3a	3a 3a	3 <i>a</i> 3 <i>a</i>	30 90	3a 3a	3 <i>a</i> 3 <i>a</i>
TRANSACTIONS OF OF		e 3)	band	$\frac{2\nu_2+\nu_3}{2\nu_1}$	$3\nu_3$	$2\nu_{2}^{\nu_{2}} + \nu_{3}^{\nu_{3}}$	$ \nu_1 + \nu_3 $ $ 2\nu_3 + \nu_3 $	$2\nu_{2} + \nu_{3}$	$2\nu_1 + \nu_3$ $2\nu_1$	$2\nu_2 + \nu_3 \\ 2\nu_5 + \nu_3$	$2\nu_2 + \nu_3$	$2\nu_2 + \nu_3 \\ 2\nu_2 + \nu_3$	$2\nu_2 + \nu_3$ $2\nu_2 + \nu_3$	$2v_2 + v_3$ $2v_2 + v_3$ $0v_3 + v_3$	$2\nu_2 + \nu_3$ $2\nu_2 + \nu_3$	$2v_2 + v_3$ $2v_2 + v_3$	$2\nu_2 + \nu_3$
	7000 FIGURE 3	TABLE 2 (figure 3) -1)	ident.	H20 H30	CO2	H ₂ O	Н20 Н,0	H20 H20	H2O	H ₂ O H ₂ O	$H_{2}^{2}O$	н20 Н,0	H ₂ O H	о П Н	H ₂ 0	∫ H ² OO	$(\odot Fe H_2O)$
Sciences and the sciences and the sciences are sciences and the sciences are sciences and the sciences are sc	2	$\nu \ (\mathrm{cm}^{-1})$		7001-9	6.0060	6963-0	6954·8 6946·9	6937-2	6923-4	6917-0 6910-1	6905-0	6871·2	6859.8 6847.7	6843.3 6843.3	6816·3	6804-3 6799-3	6791.7
Science Science		line	no.	26	100	50	31 31	32	34 8	35 36 36	37	80 00 00	40	42	44	45 46	47
			ref.	a_a	3 <i>a</i> 9 -	3a 3a	3a 3a	3a	3a	aa 3a	3a	3a 3a	3a	30 30 7	3a	3a 3a	3a 3a 3a
SOCIETY			band	$2\nu_1, \nu_1 + \nu_3$ $\nu_1 + \nu_3$	$\nu_1 + \nu_3$	$v_1 + v_3$ $v_1 + v_3$	v1+v3 v1+v3	$v_1 + v_3$	$v_1 + v_3 \\ v_1 + v_3$	$v_1 + v_3$ $v_1 + v_3$	$v_1 + v_3$	$v_1 + v_3 \\ v_1 + v_3$	$v_1 + v_3$	$2\nu_1, \nu_1 + \nu_3$ $2\nu_1, \nu_1 + \nu_3$ $9\nu, \nu, \pm \nu$	$v_1 + v_3$	1	$2 u_2 + u_3 \\ 2 u_1 \\ 2 u_2 + u_3$
CTIONS	7400		ident.	H2O H,O	H ₂ O	H ² O	H2O H3O	$\rm H_{10}^{20}$	H2O	H20 H,0	$\mathbf{H}_{2}^{2}\mathbf{O}$	H2O	H ₂ O	H ² O	H20	н СО С	H20 H200 200
		ν (cm ⁻¹)	(vac.)	7419.6 7405.3	7390-4	7327.3	$7311 \cdot 3$ $7293 \cdot 9$	7272-9	7242.9	7232·1 7219·7	7204.1	7194-6 7181-6	7167.3 7138.4	7120-0	7094.4	7054-1	7045-5 7022-5 7015-8
		line	no.	- 01	ຕ ,	₩ ν Ω (9 1-	∞ c	ی 10	11	13	14 15	16	18	88	22 22	2324







	band						1		1		1		NAME OF TAXABLE PARTY.	
	ident.	Fe S:	50 ©⊙	i Si 0 0	⊙ ⊙ Si	CH₄	• Fe	• Fe	•	CH,	Si III III III III III III III III III	• Fe	• Al	
	$ \nu (\mathrm{cm}^{-1}) $ (vac.)	6184.4	6172-2	6155.2	6127-1 6102-7	6096.4	6079.2	6063.8	6036-3	6002.5	5993.2	5979-5	5968-3	
	line no.	21	22	53	25 25	26	27	28	29	30	31	32	33	
	ref.	3a	3a	3a	3a	3a	3a	3a	3a	3 <i>a</i> 3 <i>a</i>	3a	3a 3a	3a	
re 4)	band	1		-			1	$\nu_1 + 4\nu_2 + \nu_3$	-	$\nu_1 + 4\nu_2 + \nu_3$	$\nu_1+4\nu_2+\nu_3$		$\nu_1 + 4\nu_2 + \nu_3$	
TABLE 2 (figure 4)	ident.	• Fe	⊙ Si Si	• Mg	● ● Si	• Ee	• AI	$\int CO_2$	Ee U	• CO2 Fe	$\int_{-\infty}^{\infty} CO_2$	Si Si	$\int co_2$	
TAI	$ \nu \left({ m cm}^{-1} ight) \left({ m vac.} ight) $	6320-0 6919-9	0.0100	$6292 \cdot 1$		6263.9		6946.9		6232-4	6218-7	1 1100	+. 1170	
	line no.	13	14	15		16		71	-	18	19	C C C	02	
		3 <i>a</i> 32												
	band	$v_1 + 2v_2$	$v_1 + zv_2$				1		$2\nu_1 + 2\nu_2 + \nu_3$	$\frac{-1}{2\nu_{1}+2\nu_{2}+\nu_{2}}$	° . 7			$2\nu_1 + 2\nu_2 + \nu_3$
	ident.	H_2O	● Fe	• Fe	⊙⊙ Sie	iz O O	⊙ Si	• Fe	∫ CO₂	CO°	E Fe	• Mg	• Ee	$ CO_2$
	$\nu (\mathrm{cm}^{-1})$	6605-9 650-9	00089-2 6536-4	6493.9	6435.6	6425-8	6412.0	6399-6	6370-9		6358·2	6347.8	6341.0	
					້ດ									

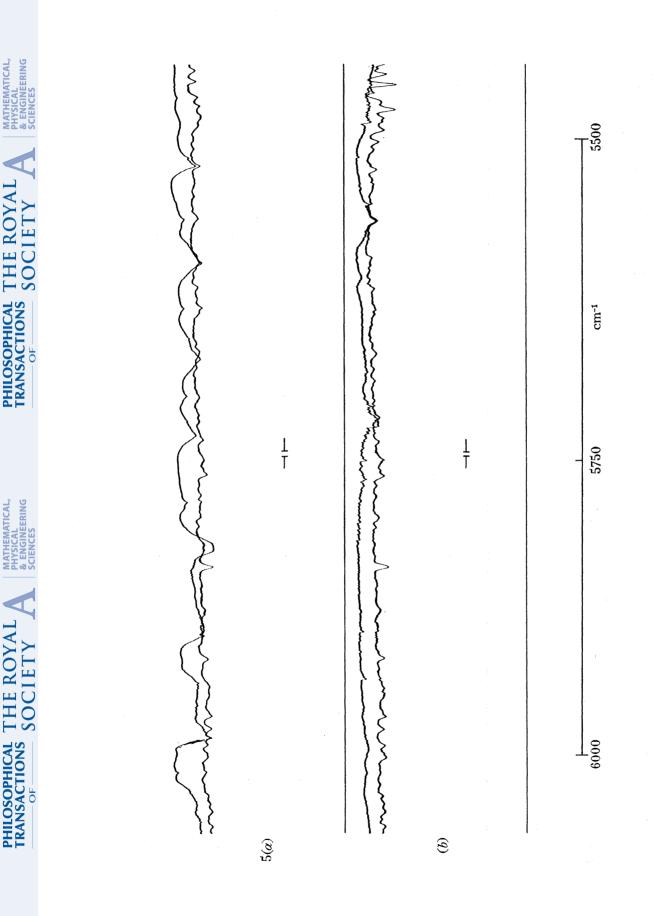
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cm⁻¹

FIGURE 4

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		н на ва
	_	band $\sum_{k=1}^{n} \sum_{k=1}^{n} \sum_{k=1}^{n$
	2200	ы ен 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1.
40		$ \nu (cm^{-1}) (vac.) (vac.) 5621.6 5662.8 55614.0 55614.0 55579.4 55579.4 55570.1 5551.8 5551.8 5551.1 5551.1 5551.1 5551.2 5551.2 5551.2 5551.2 5551.2 5551.2 5551.2 5551.2 5551.2 5551.2 5551.2 $
		line 10. 33 33 33 33 33 33 33 33 33 33 33 33 33
30	cm-1	
		band band $\nu_2 + \nu_3$ $\nu_2 + \nu_3$ $\nu_3 + \nu_3$ ν_3
	5750 FIGURE 5	$\begin{array}{c} \begin{array}{c} \begin{array}{c} \begin{array}{c} \begin{array}{c} \begin{array}{c} \begin{array}{c} \end{array} \end{array} \end{array} \end{array} \end{array} \\ \begin{array}{c} \begin{array}{c} \end{array} \end{array} \end{array} \end{array} \end{array} \\ \begin{array}{c} \begin{array}{c} \end{array} \end{array} \end{array} \end{array} \\ \begin{array}{c} \end{array} \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \end{array} \\ \begin{array}{c} \end{array} \end{array} \\ \end{array} \\ \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \end{array} \\ \end{array} \\ \end{array} \end{array} \\ \begin{array}{c} \end{array} \end{array} \\ \end{array} \\ \end{array} \\ \end{array} \end{array} \\ \end{array} \end{array} \\ \end{array} \end{array} \\ \end{array} \\ \end{array} $
	Ē	
		line 14 15 16 17 19 18 12 22 22 22 22 22 22 22 22 22 22 22 22
-		
	0	band $\nu_2^{\nu_2} + \nu_3^{\nu_2}$ $\nu_2^{\nu_2} + \nu_3^{\nu_3}$ $\nu_2^{\nu_2} + \nu_3^{\nu_3}$
(i)		ident. ident. $(\odot, Si) = CH_4$ $(\odot, Si) = CH$
5(<i>d</i>) ²	1	$ \nu \ (\text{cm}^{-1}) $ $ (\text{vac.}) $ $ (\text{vac.}) $ $ 6002.5 $ $ 5993.2 $ $ 5993.1 $ $ 5915.1 $ $ 5811.7 $ $ 5831.7 $ $ 5831.2 $ $ 5833.2 $ $ 5833.2 $ $ 5833.2 $ $ 5827.1 $ $ 58227.1 $ $ 5809.0 $
		l ان 1 - 2 - 3 1 - 3 - 4 - 4 - 3 1 - 3 - 4 - 5 1 - 5 2 - 5 1 - 5 2 - 5 1 - 5 2 - 5 2 2 - 5 2 2 2 - 5 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2

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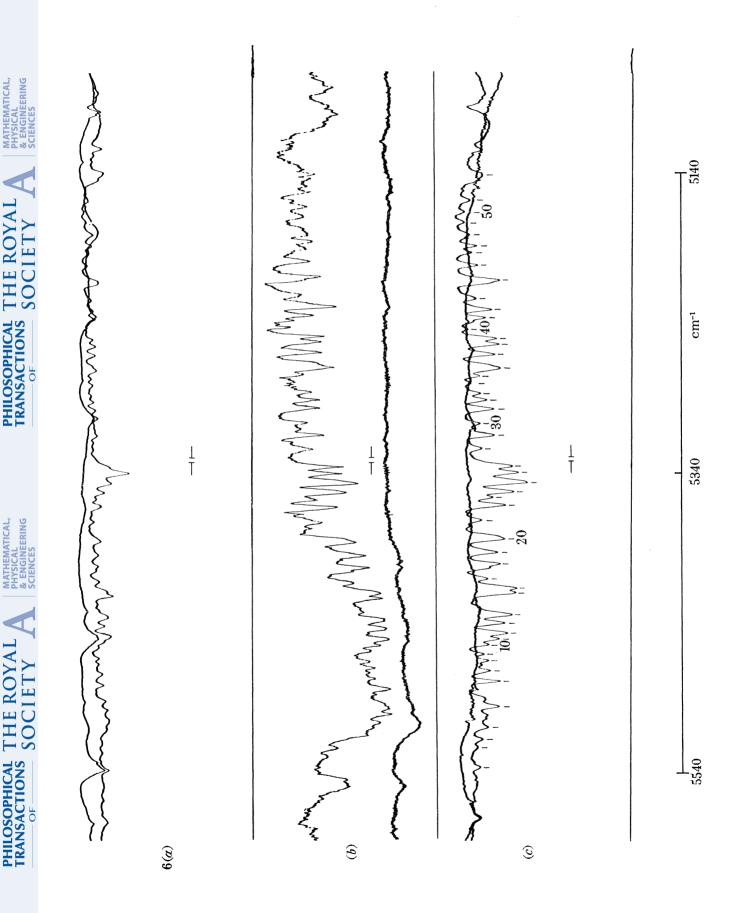
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		ref. 3 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0
		band $p_2 + p_3$ $p_2 + p_3$ $p_3 + p_3$ $p_4 + p_4$ $p_5 + p_3$ $p_5 + p_4$ $p_5 + p_5$ $p_5 + p_5$
	5140	ident.)))))))))))))))))))
		ν (cm ⁻¹) (vac.) 5265.9 5256.6 5226.1 5226.1 5226.1 5229.9 5220.3 5220.3 5220.3 5220.3 5220.3 5129.0 5189.0 5189.0 5181.9 5157.1 5157.2 517.2 51
ζ		line 337 510 52 53 53 53 54 54 54 54 55 55 55 55 55 55 55 55 55
	cm ⁻¹	$\begin{smallmatrix} \mathrm{ref}_{a} \\ \mathrm{sec} \\ se$
		e 6) band $p_{2}^{2} + p_{3}^{2}$ $p_{2}^{2} + p_{3}^{2}$ $p_{3}^{2} + p_{3}^{2} + p_{3}^{2}$ $p_{3}^{2} + p_{3}^{2} + p_{$
	1 5340 Figure 6	TABLE 2 (figure 6) TABLE 2 (figure 6) 1 ident. 1
		$\begin{array}{c} {\rm TA}_{\rm A} \\ \nu ({\rm cm}^{-1}) \\ \nu ({\rm vac}.) \\ 53860 \\ 53860 \\ 53860 \\ 53860 \\ 53364 \\ 53364 \\ 53394 \\ 53394 \\ 53394 \\ 53394 \\ 53394 \\ 53394 \\ 53394 \\ 53394 \\ 53294 \\ 53294 \\ 53292 \\ 532$
		line 10. 22 22 23 23 23 24 20 25 23 20 26 25 22 20 20 20 20 20 20 20 20 20 20 20 20
		ref.
		band $\nu_2^{\nu_2} + \nu_3^{\nu_2}$ $\nu_2^{\nu_2} + \nu_3^{\nu_2}$ $\nu_2^{\nu_2} + \nu_3^{\nu_3}$ $\nu_2^{\nu_2} + \nu_3^{\nu_3}$
	5540	існи прососососососососососососососососососос
و (<i>đ</i>) بر		ν (cm ⁻¹) (vac.) 5536·1 5521·1 5521·1 5516·0 55495·8 54495·8 5449·5 5448·0 5548·1 5480·1 5480·1 5548·1 5480·1 5548·1 5480·1 5480·1 5480·1 5480·1 55480·1 5480·1 5480·1 5480·1 5480·1 5480·1 5480·1 5540·0 5480·1 5540·1 5560·1 5560·1 5560·1 5560·1 5560·1 5560·1 5560·1
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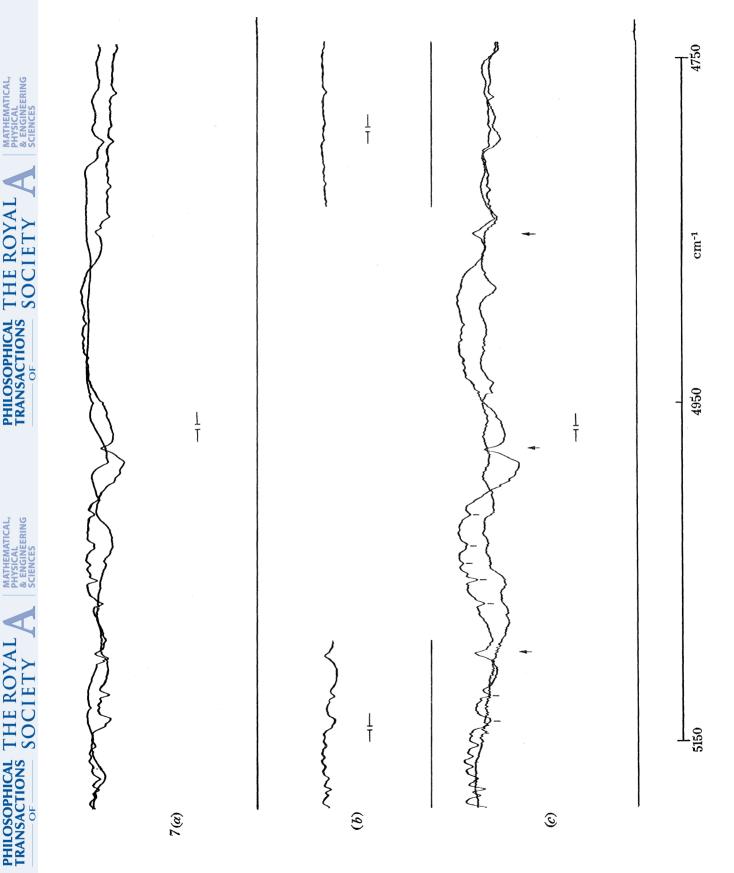
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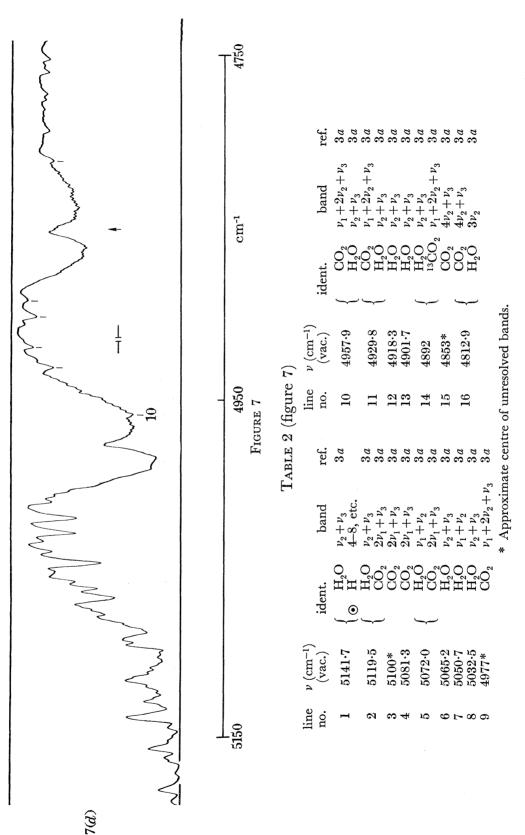
ATLAS OF THE INFRA-RED SOLAR SPECTRUM

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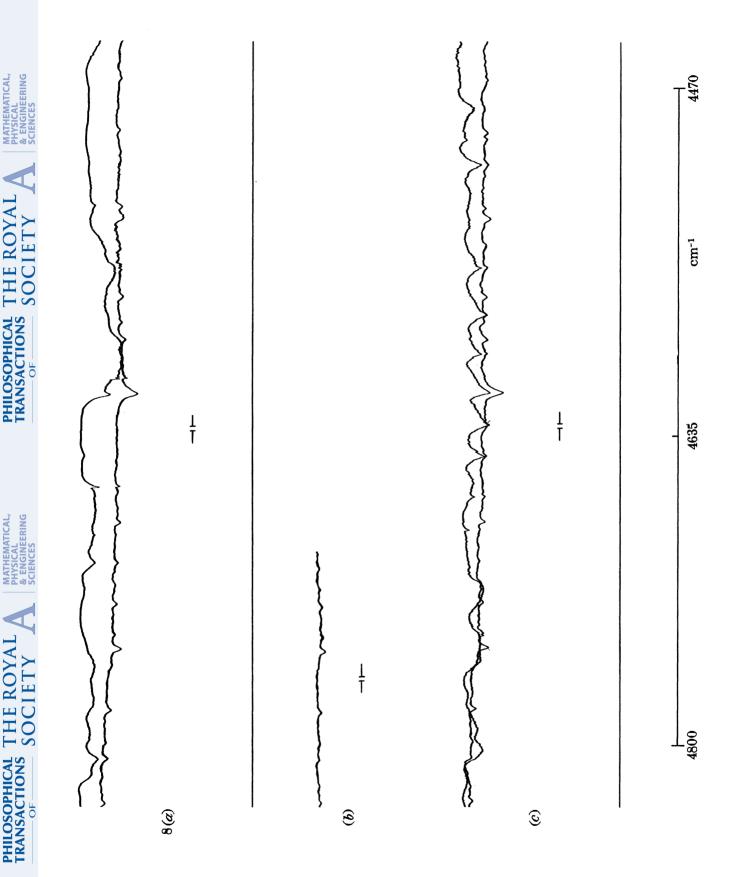
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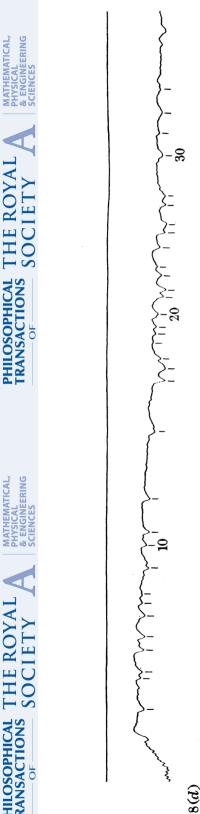
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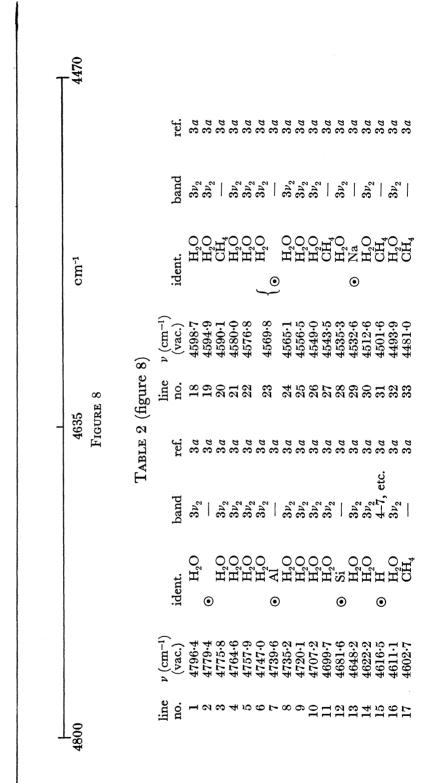
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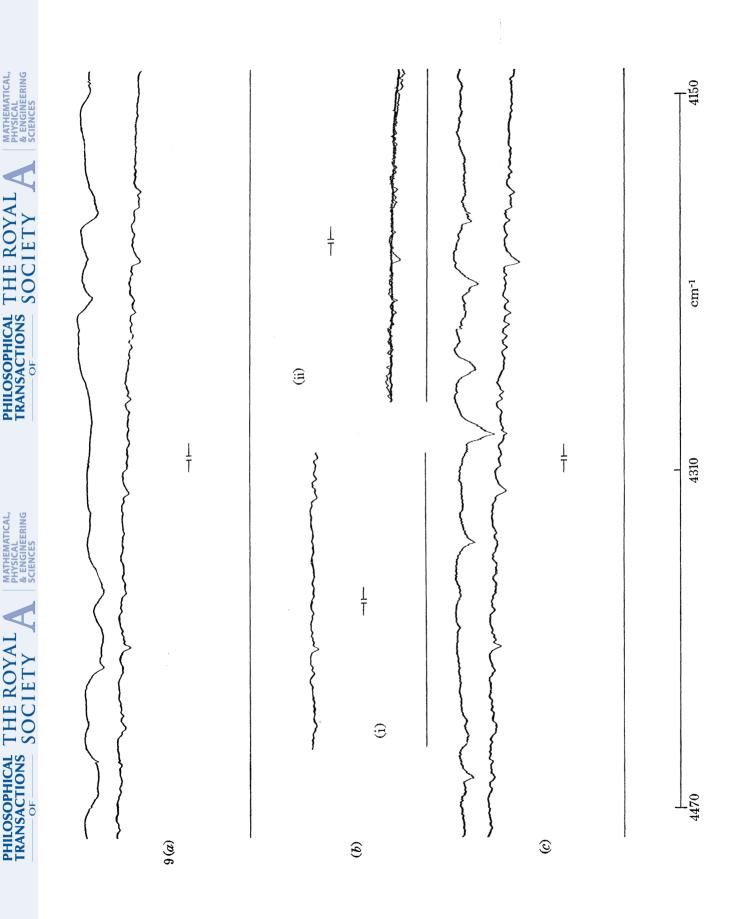


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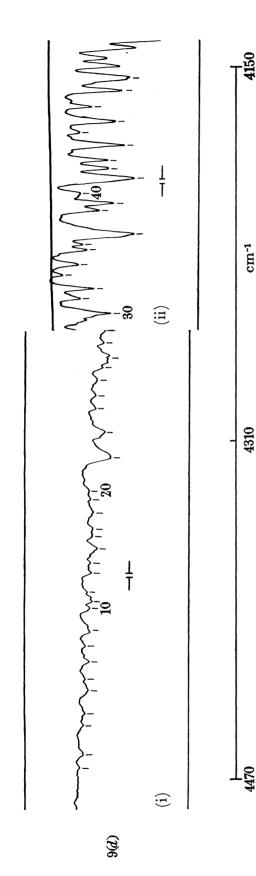








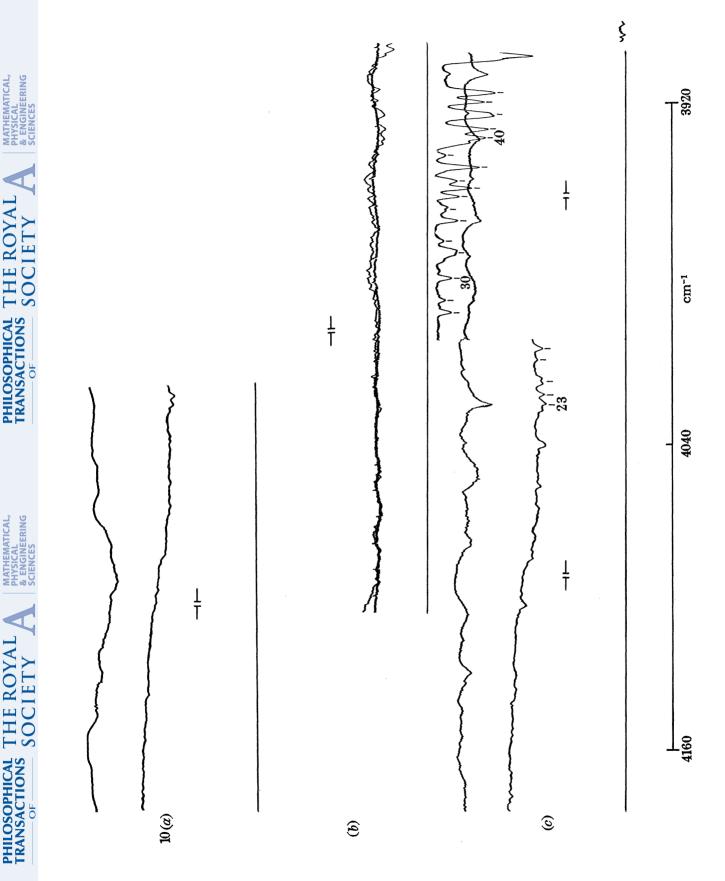




			-																		
			band	P3	٣,	ۍ م	י	\mathcal{V}_{2}	ۍ م	ب بع	, <i>v</i>	ν. γ,	بر بر	ν. •	ب بع	٣,	٣,	, بر ور	<i>ب</i> ر .	, ^{بر}	5
			ident.	Н,О	H,O	H,O	ſ CŤI₄	Ó,H	H,O	H,O	H ₀ H	H,O	H ₀	H _. O	H ₀ ,H	H,O	H,O	H ₀ H	H,O	H ₀ ,H	1
		$\nu \; ({\rm cm}^{-1})$	(vac.)	$4230 \cdot 1$	4224.3	4221.2	0.0101	0.0177	4208.0	4204.8	4200.2	4194.5	$4191 \cdot 1$	4187.3	4181.5	4176.4	4171-3	4165.9	4159.2	4154.6	
		line	no.	34	35	36	L0	10	38	39	40	41	42	43	44	45	46	47	48	49	
			ref.	3a	3a	3a	3a	3a	3a	3a	3a	3a	3a	3a	3a	3a	3a	3a	3a	3a	3a
	; 9)		band	1		1		$ \nu_3 $	ν_3	ν_3	1			ν_3	ν_3	ν_3		ν_3		ν_3	
FIGURE 9	TABLE 2 (figure 9)		ident.	\int H ₂ O	(CH₄	CH ⁴	CH ⁴	0,H	$H_2^{-}O$	H_2O			∫⊙ Na		H_2O	$H_{2}^{-}O$	CH_4	$\rm H_2O$	CH_4	H_2O	CH_4
	T_{AB}	$\nu \; (\mathrm{cm}^{-1})$	(vac.)	1212.0	B.0707	4333.7	4315.6	4306.7	4294.5	4288.8	7.1864	LIOTE	1976.1		4269.8	4264.3	4261.1	4250.8	4244.8	4239.9	4234-4
		line	no.	10	гa	20	21	22	23	24	95	3	96	04	27	28	29	30	31	32	33
			ref.	3a	3a	3a	3a	3a	3a	3a	3a	3a	3a	3a	3a	3a	3a	3a	3a	3a	3a
			band					1			1	ν_3]			ν_3
			ident.	CH_4	CH ¹	CH ¹	CH ¹	CH₁	CH ¹	CH_{4}	CH_{4}	H_2O	CH_4	CH ⁴	CH_{4}	CH4	CH_{4}	CH_{4}	CH4	CH_{4}	$\rm H_2O$
		$\nu \ (\mathrm{cm}^{-1})$	(vac.)	4466.9	4463.7	4456.3	4449.9	4435.8	4430.8	4421.4	4415.5	4407.8	4400.5	4394.3	4391.3	4384·3	4380.6	4375.6	4367.0	4362.0	4356·I
		line	no.		2	က	4	ñ	9	7	x	6	10	11	12	13	14	15	16	17	18

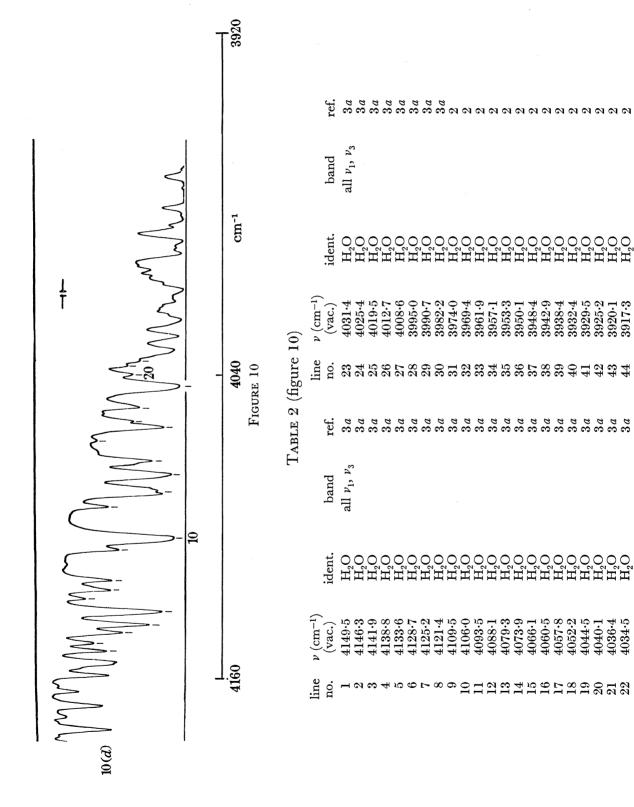
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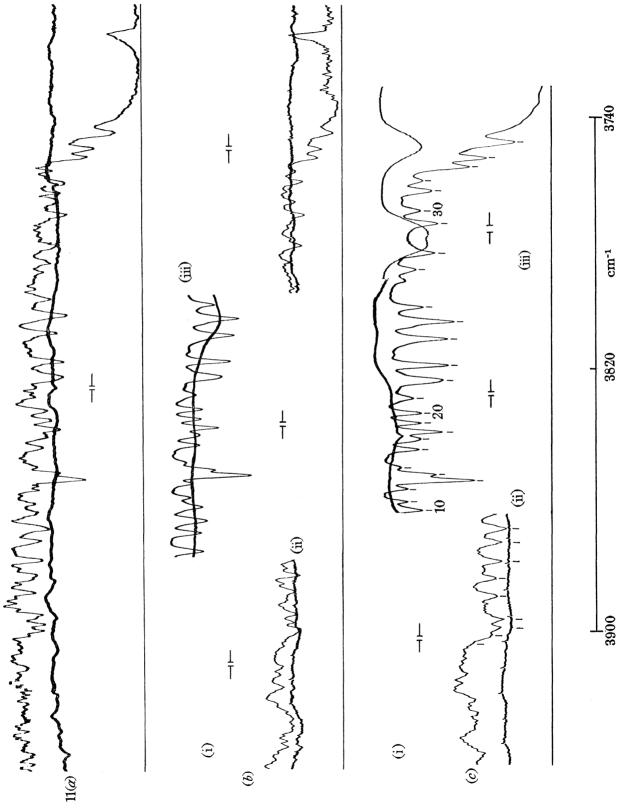














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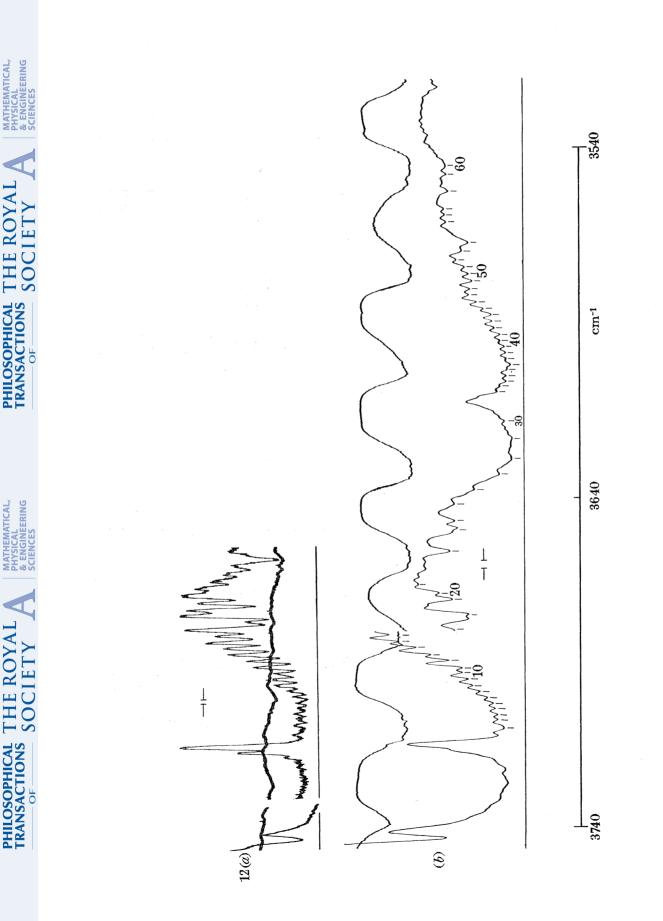
Residual water vapour in spectrometer at 45000 ft.

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40			ref.	¢	10	1 01	61	ଧ	61		5	67	01	61	61	01	5	01	5	21	67
3740			band	2	رس بر	د . 3	<i>ب</i> ر .	<i>ب</i> ر. 23	ν_3	4-6, etc.	ν_3	ν_3	ν_1, ν_3	ν_3	ν_1, ν_3	ν_3	ν_1, ν_3	ν_3	ν_1, ν_3	ν_1, ν_3	ν_3
cm ⁻¹			ident.	Он	H ₀	H,0	H,O	H,O	$\int H_2^{-}O$	H O	H_2O	H_2O	$H_2^{-}O$	H_2O	$\rm H_2O$	H_2O	$H_{2}^{-}O$	H ₀	H _. O	$H_{2}O$	$H_2^{-}O$
		[1]	$\nu (\mathrm{cm}^{-1})$	2825.0	3831.7	3826.7	3821.7	3816.0	3806.7	-0000	3801.2	3796.1	3784-4	3779.4	3769-7	$3765 \cdot 6$	3759-8	3756 - 6	$3752 \cdot 1$	3749.3	37 44 ·0
3820	FIGURE 11	(figure	line no.	10	06 06	21	22	73	76	H I	25	26	27	28	29	30	31	32	33	34	35
	Figu	TABLE 2 (figure	ref.	6	1 C.	101	67	61	67	61	61	67	01	61	61	01	01	01	67	c)	63
		I	hand	-	2 ² 3	V3. V3	V. 5	v_{1}, v_{3}	ν_3	ν_3	ν_1, ν_3	¥.3	<i>ب</i> ر 2	ν_3	ν_3	ν_3	<u>ب</u> ر	ν_1, ν_3	۳.	ج چ	ν_3
			ident.	Он	H_O	H,O	H,O	H,O	H_2O	$H_{2}^{-}O$	$H_{2}^{-}O$	$H_{\overline{0}}$	$H_{2}^{-}O$	H_2O	$H_{2}O$	H_O	H ₀	H ₀	H,O	H,O	$H_{2}^{-}O$
3900			$\nu (\mathrm{cm}^{-1})$	2006.9	3904-9	3901-9	3899.4	3891.3	3886.0	3880.3	3874.5	3870.0	$3865 \cdot 2$	3861.9	3857-4	3854.3	3852.3	3849.9	3843.9	3841.2	3838-0
			line.	-	- 6	100	4	ŝ	9	2	æ	6	10	11	12	13	14	15	16	17	18





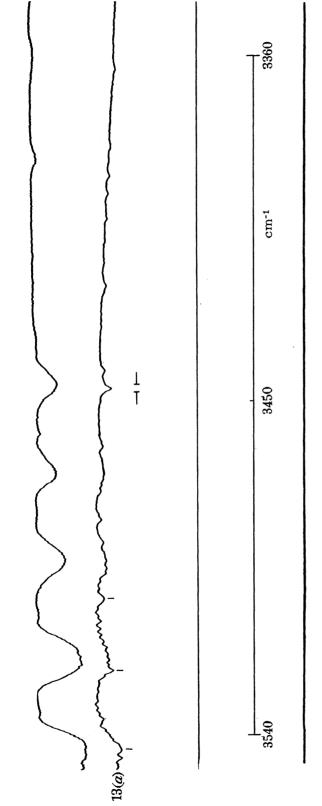
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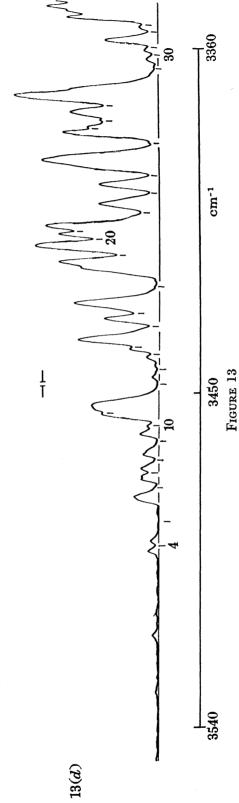




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~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~						P26 P28	P30	P32	P34 P36	P38	P40	P42	P44 $P46$	DAQ	0# 1	P50						
ł					band	+ + v3	r + 3 	$+\nu_{3}$	$2\nu_2 + \nu_3$ $2\nu_5 + \nu_5$	- +	$+\nu_3$	$+\nu_3$	ہ ہے 1 + 4	° :	₽3 ₽3	v, + v,	$+ \nu_3$	v3 + v3	$v_1^{\nu_1}$	- 	2	ب م
1						$2\nu_2$	$2v_{2}^{V}$	5°2 10	8 7 7 7 7	50	2,2	$5^{-5}$	8 7 2 2 2	0.3° 2	V1'	2 ^v 2	$2v_2$	$2^{\nu}_{\nu}$	, ¹ 6	~ ¹	$\nu_3$	<b>1</b>
		3540			ident.	ဝိုင်	$\left\{ \begin{matrix} \mathrm{CO}_{2}^{2} \\ \mathrm{H}_{2}^{2} \mathrm{O}_{2} \end{matrix}  ight\}$	CO2	o o o co	$CO_2^2$	CO ² H.O ²	CO2	ççç ççç	$\{ \mathbf{H}_{\mathbf{O}}^{\mathbf{O}} \}$	H ₂ O ²	(CO ₂ )	$\{ CO_2 \}$	(H20) (C0)	H ² Ó	$H_2^{02}$	H ₀	++2~
					$     \nu  (\mathrm{cm}^{-1}) $ (vac.)	3590-6 3588-7	3586.9	3585.2	3583·3 3581·3	3579.4	3577-4	3575.1	3572.8 3570.5	3568-0	3566-0	0 0020	3002.3	3560.1	3557.1	3552·3	3547.2 3545.2	1 0100
					line no.	$\frac{42}{43}$	44	45	46 47	48	49	50	51 52	53	54	ւ	66	56	57	58	59 60	3
		cm ⁻¹			ref.	20	ର ର															
		Ŭ			pq				$^{-}_{R30}$		3 R20	3 R10	R6			$^{3}_{P10}$			3 P16			•
			12	gure 12)	band	بر بر م	د ۳, ۳3 ۳,	C	$2\nu_{2} + \nu_{3}$ $2\nu_{6} + \nu_{6}$	$\nu_3$	$2\nu_2 + \nu_3$ $\nu_5$	$2\nu_2 + \nu$	$v_3^{\nu_3}$	$\nu_1, \nu_3$	$2v_2 + v_3 + v_3$	$2\nu_2 + \nu$ $2\nu_3 + \nu$	$2\nu_2^2 + \nu$	$2\nu_2 + \nu_1$	$2\nu_2 + \nu_3$ $2\nu_2 + \nu_3$	$2\nu_2 + \nu$	$2\nu_2 + \nu$ $2\nu_2 + \nu$	122 -
	-	3640	FIGURE 12	TABLE 2 (figure 12)	ident.	H20 H30	H ₂ O H ₃ O	$\{ \overset{H_2}{\operatorname{H}_2^2} O \}$		$(H_2 \vec{O})$	CO ₂ H,O		(H ² O)	$\{H_2^{O}$	000000000000000000000000000000000000000	S O	CO2	(CU ₂ (H,O			o°c Coc	< (2 (2
		(1)		$T_{Al}$	$\nu  (\mathrm{cm}^{-1}) $ (vac.)	$3656 \cdot 5$ $3649 \cdot 2$	3647.0 3642.0	3638-0		3634-0	3628-3	$3621 \cdot 2$		3015-U 3600.6	3608·1	3606.5 3604.9	3603.1	3601.4	3599-7 3598-2	3596.2	3594.5 $3592.7$	-
					line no.	$22 \\ 23 \\ 23 \\ 23 \\ 23 \\ 23 \\ 23 \\ 23 \\$	$24 \\ 25$	26	Ċ	12	28	29	0	30 31	32	33 34 34	35	36	37 38	39 39	40 41	1
					ref.	ର ର	ରା ଚା	01 (	20	0	ରା ରା	010	2 2	010	2010	20	c1 (	20	01 O	101	S) S	। <b>२</b> १
					band	$\begin{array}{c}\nu_1+\nu_3  P4\\\nu_1\end{array}$			$v_1 + v_3 P10$ $v_1 + v_5 P12$					$v_1 + v_3 P24$				$     v_1 + v_3  F34 \\     v_1 + v_3  P36 $		$\nu_1 + \nu_3$	$\nu_3$	<u>د</u> ا
		0			ident.	$\{ \substack{\text{CO}_2\\\text{H},\text{O}_2}$		(H ₂ Ó	ဝိုင်	$\operatorname{CO}_2^{\scriptscriptstyle 2}$	(CO ₂ (H,O	Ç02	ဝိုင်			(U0, (H,0	CÕ2	çç CC	CO ² (H ² O	$\{\overline{CO_2}^{2}\}$	H2O H2O	) 7
<b>Weinigen and Annual Sector</b>	- <b></b>	3740			$     \nu  (\mathrm{cm}^{-1}) $ (vac.)	3711.4		1.2015	3704-7 3704-7	$3703 \cdot 0$	3701.5	3699.5	3696-0 3696-0	3694•1 3692•3	3690-7	3688.3	3686.5	3684•5 3682•6	3680-5	3675.0	3670-9 3668-8	3661·2
					line no.	Ι	G	N 6	ю 4	ю,	9	<b>L</b> 0	xo	10	12	13	14	15 16	17	18	19	21



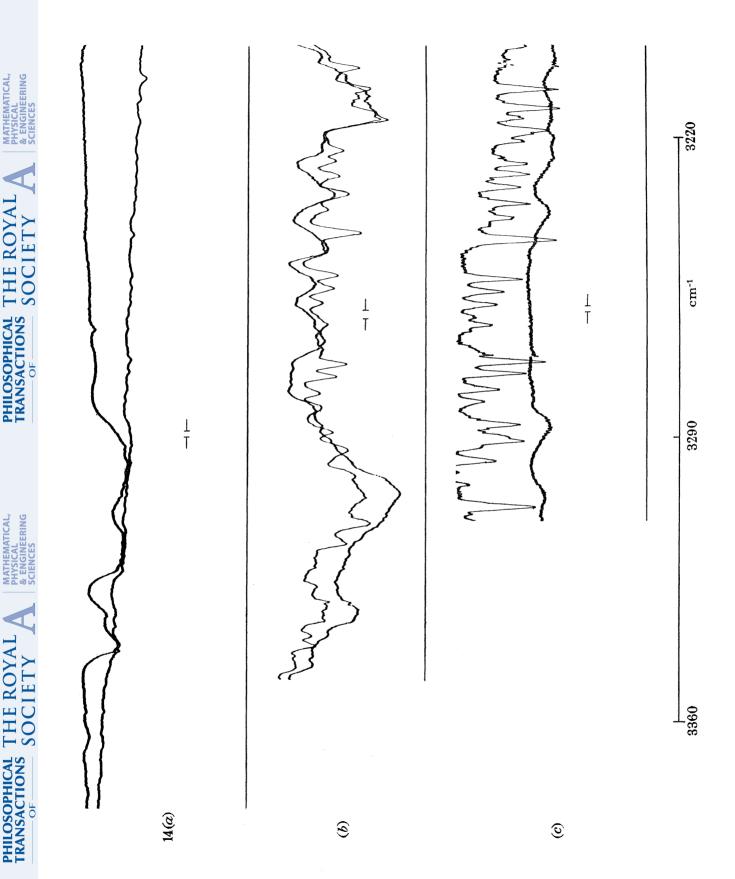




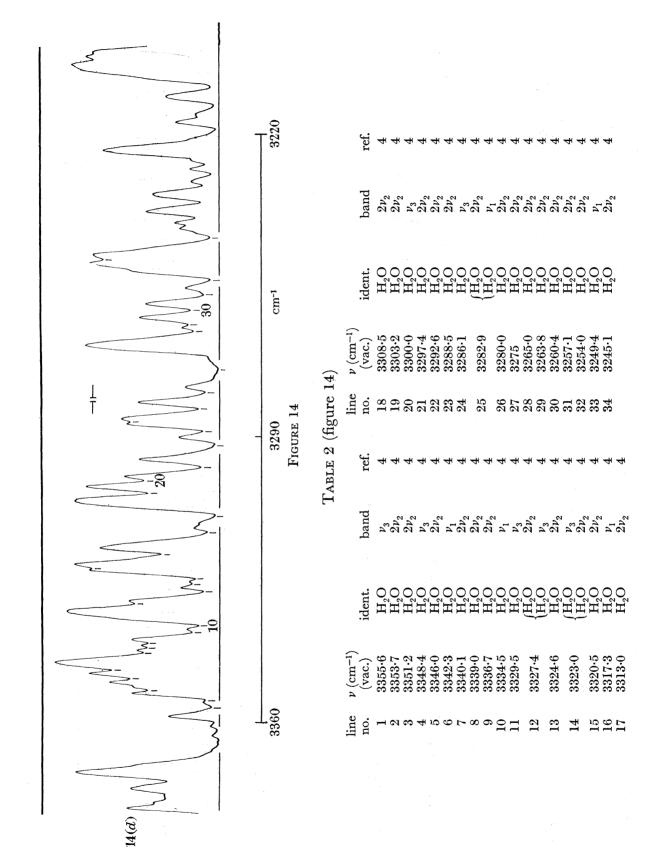
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	band	ν ₃ ν1	733 A	⁷ 2	$2\nu_2, \nu_1$	$\nu_{1}, 2\nu_{2}$	$2\nu_{,}$	<i>ب</i> اً۔	$2\nu_{s}$	ν, -	$2\nu_{s}$	<i>ν</i> , ²	ν.,	$\nu_3$	2v,	1
TABLE 2 (figure 13)	ident.	$_{ m H_{0}^{2}O}^{ m H_{2}O}$	H ₂ O H	$H_2^{20}$	H2O H3O	$H_{0}^{2}O$	$H_{0}$	$H_{0}$	$H_{0}$	$H_{0}$	H,O	H,O	H,O	H,O	H,O	a
	$\nu (\mathrm{cm}^{-1})$ (vac.)	3427.9 3420.4	3413.0 3408.8	3406.7	3403.6 3397.2	3392.7	$3385 \cdot 6$	3380-4	3377-5	3374.7	3365-7	$3361 \cdot 6$	3359-5	3355.6	3353-7	
	line no.	17 18	19	21	33 K	24	25	26	27	28	29	30	31	32	33	
	ref.	07 4	44	4	44	4	4	4	4	4	4	4	4	4	4	4
	band	$v_1, v_3 2v_3 + v_3$	$v_{3}^{\nu_{3}}$	$\nu_1$	ע ₁ ע ₃ , ע	$\nu_1, \nu_3$	$\nu_{3}, \nu_{1}$	$\nu_1$	$2\nu_2, \nu_3$	$\nu_3^{-2}$	<i>ل</i> ري 2	ν,	μ ¹	$2^{\nu_2, \nu_3}$		ν, Γ
	ident.	(H ₂ O) (13ČO,	$\left( \begin{array}{c} H_{2}O \\ 13CO_{n} \end{array} \right)$	$H_2O^2$	H2O H,O	${ m H}_2^{ m i}{ m O}$	$H_2O$	$H_{2}O$	$H_2O$	$H_{2}O$	$H_{0}$	$H_{2}^{-}O$	$H_{\overline{0}}$	$H_{2}^{-}O$	$H_{\overline{0}}$	$H_2\overline{O}$
	$\nu (\mathrm{cm}^{-1})$ (vac.)	3545-2	3528-0	3503-0	3496.7 3488.2	3474.8	3470.5	3467.2	3461.4	3455-7	3453.2	3446.6	3442.3	3440.1	$3438 \cdot 1$	3430.9
	line no.	I	5		4 v	9	2	x	6	10	11	12	13	14	15	16

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## ATLAS OF THE INFRA-RED SOLAR SPECTRUM



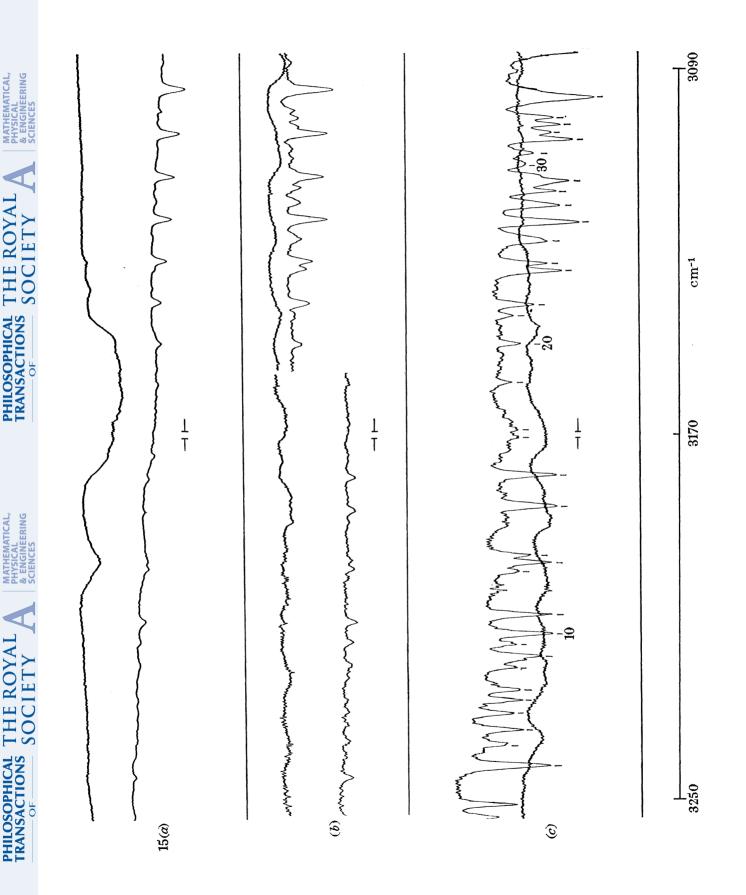
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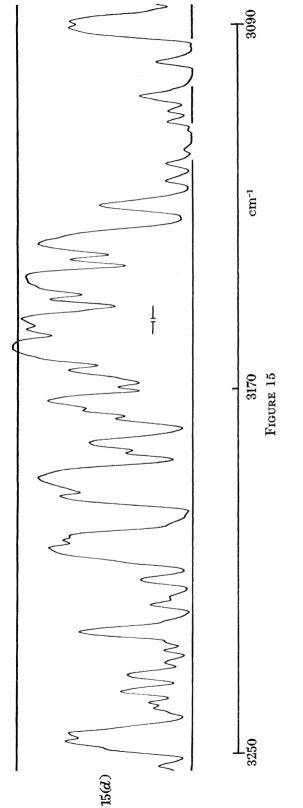
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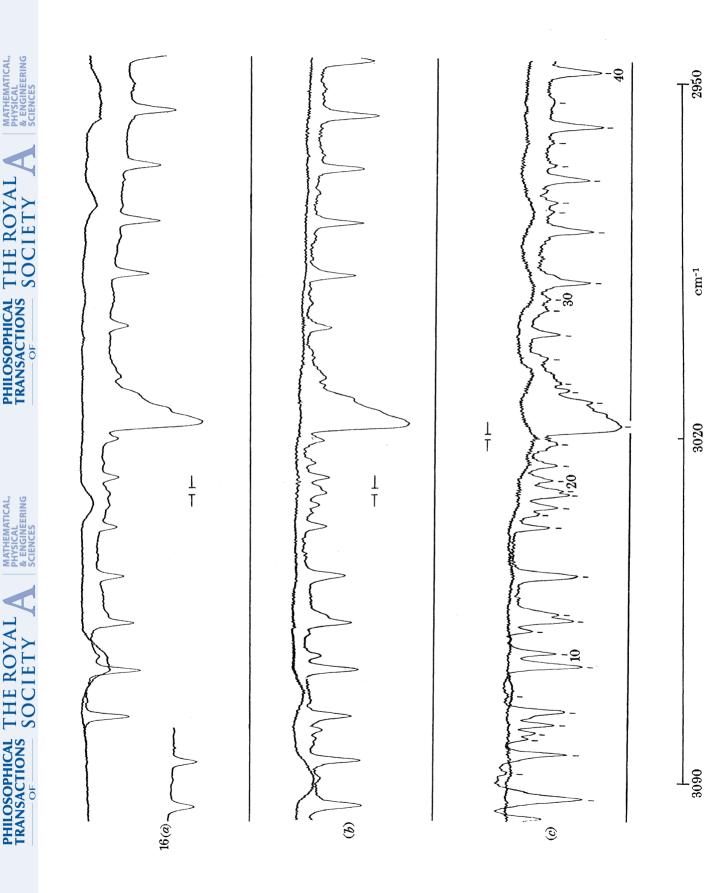
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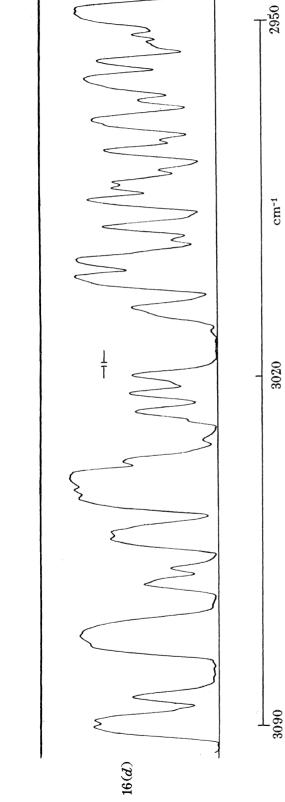
	ref.	4	4	4	4	4	4	4	4	4	4	4	4	4	4	4	4	4	4	4	4	4
	band	v. R12	$2^{j}_{\nu}$ ,	$\nu_3^{-} R11$	$2\nu_2$	.	$\nu_3  R10$	$2\nu_{s}$	$2\nu_2$	.	$2 u_{2}$	$\nu_3 R9$	$2\tilde{\nu}_2$	$2\nu_2$	$2\nu_2$	$\nu_3$ R8	$2\nu_{s}$	$2\nu_2$	$2\nu_2$		$\nu_3 R7$	$2 u_2$
	ident.	CH,	$\mathrm{H}_{0}$	CĤ₄	$/\mathrm{H_2\dot{O}}$	(CH₄	CH4	(H,O	(H ₂ O	(CH₄	Ó,H	∫CH₄	$(H_{a}O$	$H_2^{-}O$	$H_{2}^{-}O$	$CH_4$	Ó,H	$H_{O}$	$(H_{2}^{-}O)$	(CH₄	∫CH4	$(H_2O$
	$\nu (\mathrm{cm}^{-1})$ (vac.)																				2005.9	1 0000
ure 15)	line no.	22	23	24	96	0.4	96	04	10	17	28	00	63	30	31	32	33	34	26	00	36	2
TABLE 2 (figu	ref.	4	4	4	4	4	4	4	4	4	4	4	4	4	4	4	4	4	4	4	4	4
	pc																			R14	R13	
	baı	$2\nu_{c}$	$2\nu_{s}$	$2\nu_{s}$	$2\nu_2^{-}$	$2\nu_2$	$2\nu_2$	2٧,	$2\nu_{s}$	2, 2	$2\nu_{s}$	$2\nu_{s}$	$2\nu_2$	$2\nu_2$	$2\nu_2$	$2\nu_{s}$	2٧,	$2\nu_{s}$	$2\nu_{2}$	$\nu_3$	<i>ب</i> ع	$2\nu_2$
	ident.	H°O	H,O	H,Ô	$H_{0}$	$H_{O}$	$H_{0}$	H,O	H,O	H,O	H,O	H,O	H,O	$H_{0}$	$H_{0}$	H _. O	H,O	H,O	H,O	$\mathrm{CH}_4$	CH ⁴	${ m H_2O}$
	$\nu (\mathrm{cm}^{-1})$ (vac.)	3245.1	3240.0	3236.6	3232.9	3230.0	3227.5	$3222 \cdot 0$	3219.3	3216.5	3214-1	3209.8	3199.8	3197.9	3196.2	$3185 \cdot 2$	3178.2	3169.7	3167.9	3157.5	3148.9	3142.8
	line no.	_	01	ෆ	4	5	9	2	×	6	10	11	12	13	14	15	16	17	18	19	20	21



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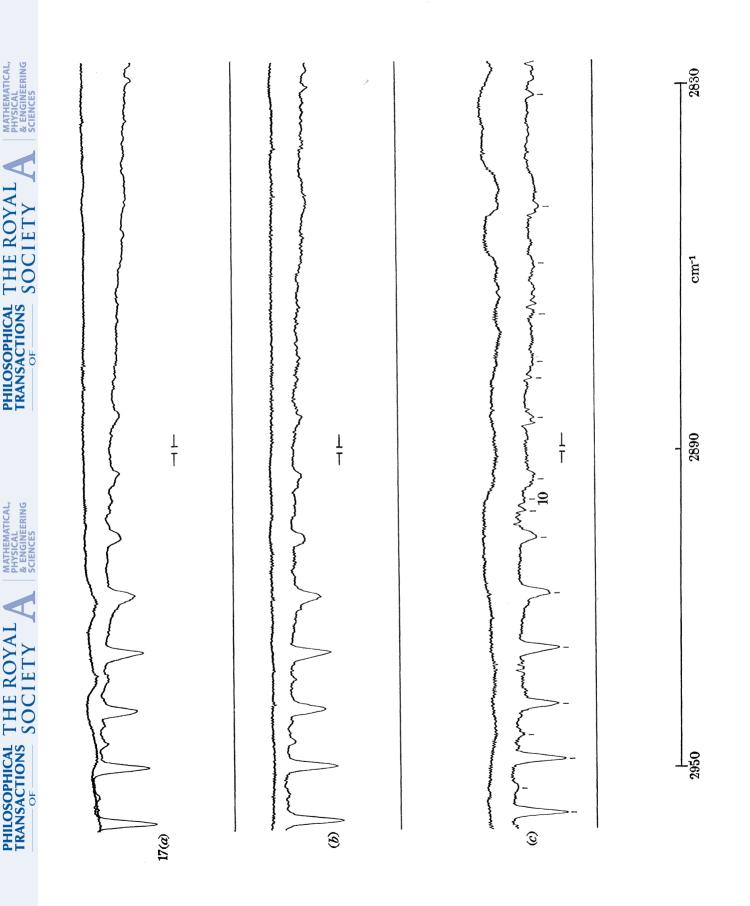


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		band	$2 u_2$	$2\nu_2^{}$ no	V3 P3	$\stackrel{\nu_3}{_{2\nu_3}}P4$	$2\nu_2^-$	$2\nu_2$	$\tilde{\nu_3}$ P5	$2\nu_2$	${2 u_2\over u_3}P6$	v3 P7	5
		ident.	$\left\{ {{\rm H}_{\rm A}^{\rm O}} \right\}$	H ₂ Ô	$H_{0}^{CH_{4}}$	(CH₄ {H₀O	$\left\{ {{\rm H}_{2}^{\rm i}{\rm O}}  ight\}$	(H₂Ô {CH,	CH ⁴	$\left\{ {{\rm CH}_{4}^{00} { m CH}_{4}^{00} } \right\}$	H2Ó CH,	$\operatorname{CH}_{4}^{1}$	M
		$     \nu  (\mathrm{cm}^{-1}) $ (vac.)	2994.4	2991.9	2988.8	2978-9	2975-2	2973-3	2968.6	2966-0	2961.5 2958.3	2953.6 2947.9	
		line no.	29	30	31	32	33	34	35	36	37 38	$\frac{39}{40}$	
		ref.	44	4,	44	44	44	44	4	44	44	44	4
16	ure 16)	band	$\nu_3 R 1 2 \nu_6$		$2v_2$ $2v_3$	$2\nu_2^{}$ $\nu_3^{}$ $R0$	$2\dot{\nu}_2$	$2\nu_2$	$ u_3  Q0$	$2^{\nu_{3}}_{\nu_{3}}$	$\begin{array}{c} 2 \nu_2^{ ilde{r}} \  u_{ m s} & 0 \end{array}$	$\sum_{v_3}^{v_3} \tilde{P1}$	$\nu_3$ ² P2
FIGURE 16	TABLE 2 (figure 16)	ident.	$CH_4$ $H_0$ O	{CH ₄	(H2O H,0	H ₂ O CH,	${\rm (H_2^{\dot{O}})}$	$(H_2^{\rm H})$	CH4	${\rm CH}_{\rm H,0}$	(H ₂ O {CH.	CH4 H,O	Cℋ₄
	$T_{A1}$	$   \nu \left( { m cm}^{-1} \right)   \left( { m vac.} \right) $	3038.5 3035.8	3034.4	3031.9	3030-8 3028-8	3025.8	3021.1	3017.3	3012-4	3010-3	3008.9 3003.7	2998-9
		line no.	15 16	21	18	$\begin{array}{c} 19\\ 20 \end{array}$	21	22	23	24	25	26 27	28
		ref.	44	4	44	44	44	44	4	44	44	44	44
		band										$\begin{array}{cc} \nu_3 & R3 \\ 2  u_5 & 3  u_5 \end{array}$	
		ident.	${\rm CH}_4 {\rm H}_2 {\rm O}$	CH4	{CH₄ (H _o O	H ₂ O	$\{ {\rm CH}_4 \ {\rm H,O} \}$	CH4 HCO	CH4	CH₄ H,O	$\left\{ {{ m H}_{{ m CH}}^2{ m O}}  ight\}$	CH4 CH4 H,O, O,	$\{ \begin{array}{c} \widetilde{CH}_4 \\ H_2 O \end{array} \}$
		$\nu \left( \mathrm{cm}^{-1} \right) \left( \mathrm{vac.} \right)$	3095.2	3089-7	3086-0	3082.6 3081.3	3079-6	3076-7	3073-3	3067.2 3064.3	3059-9	3057.7 3056.4	3048.4
		line no.	1	67	က	4 v:	9	7	ø	9	11	12 13	14

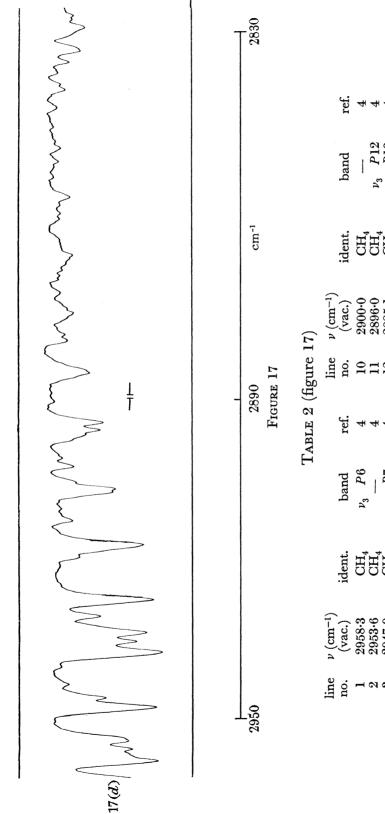
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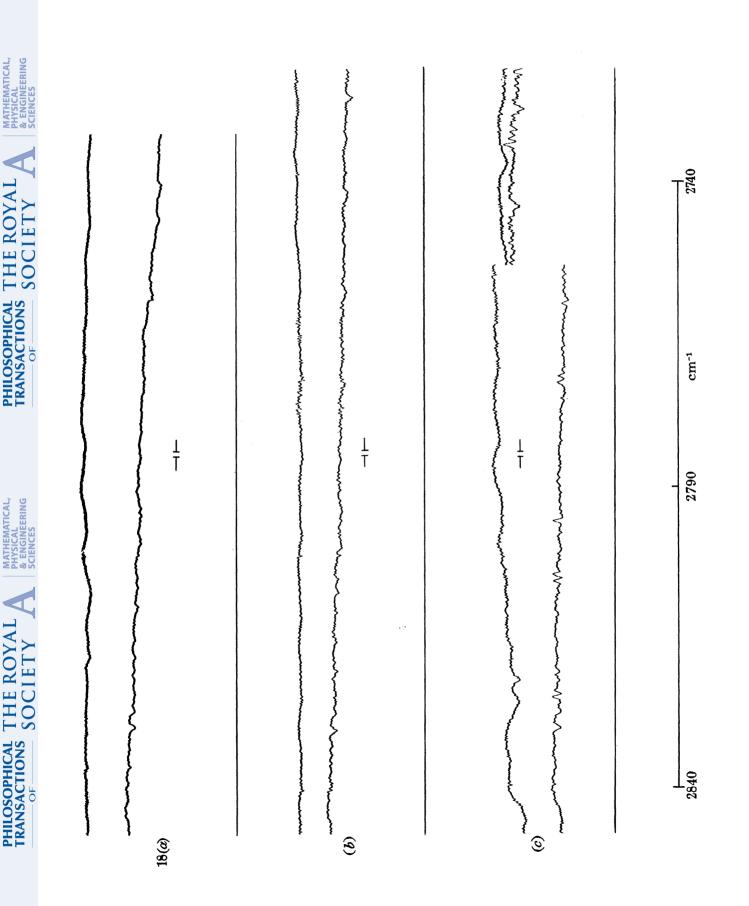
 ${\scriptstyle \begin{array}{ccc} \nu_{3} & P12 \\ \nu_{3} & P13 \end{array}}$  $egin{array}{cccc} v_3 & P14 \\ v_3 & P15 \\ v_3 & P16 \\ v_3 & P17 \\ 2v_2 & P17 \end{array}$ ident, GH4 (H₁O) (H₁O) (H₂O)  $\nu \left( {{\rm cm}^{-1}} \right) \ \left( {{\rm vac.}} \right) \ \left( {{\rm$ 2841.7 2830-8 18 17 4 44 4 4 P10P11 $\begin{array}{ccc} v_{3} & P8 \\ v_{3} & P9 \\ v_{3} & P10 \\ v_{3} & P11 \end{array}$  $\nu_3 P7$  $2\nu_2$  $\stackrel{ident}{\underset{i=1}{\overset{i}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\atopi=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset{i=1}{\underset$  $\nu$  (cm⁻¹) (vac.) 2958.3 2953.6 2947.9 2947.9 2943.3 2943.3 2943.3 2947.9 2947.9 2947.9 2947.9 2947.9 2906.7 2906.7 2901.8 -264201-8 0

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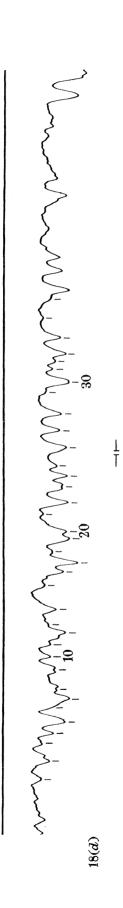
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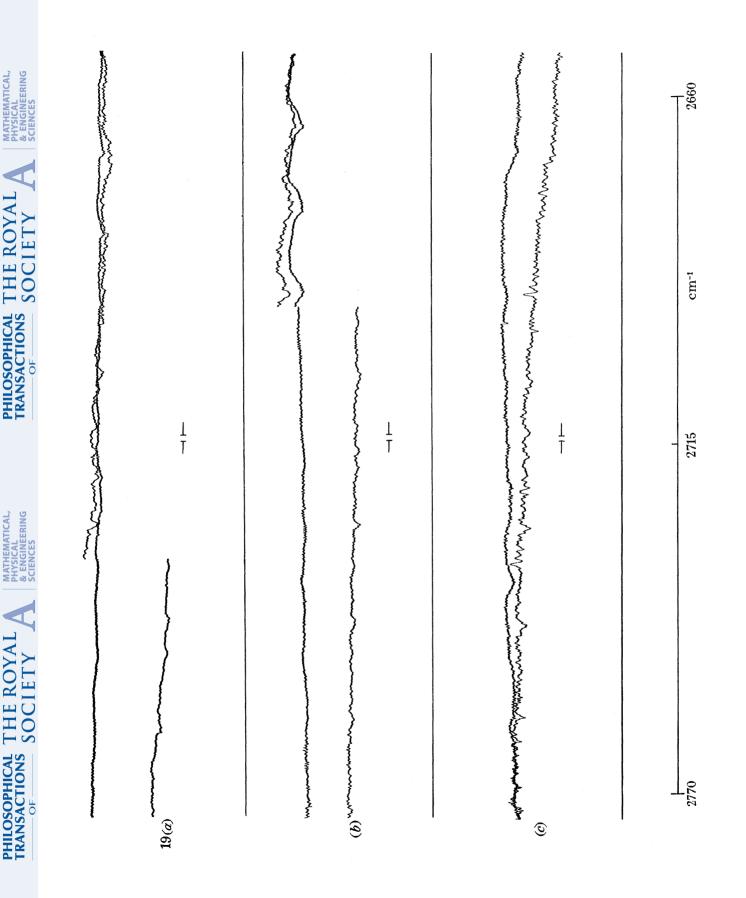


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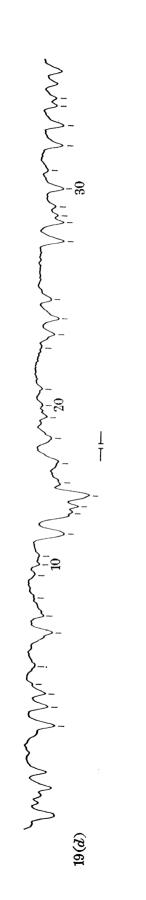
					band	ν ₁	$\nu^{}_{1}$		-	$\nu_{\rm l}$	$\nu_1$		7.7	-	ν,	۳ <u>1</u>	$\nu_{\rm I}$		$2 u_4$
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-	2740			$\nu (\mathrm{cm}^{-1})$	(vac.)	2785-9	2783.3		2.180.2	2777.5	2774.0	9779.4		1.07.12	2768.8	2767.5	2764.8	2762.0	2758-2
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	1				ref.	∞ ∞	∞ -	4 ∞	8	8	<b>∞</b> 0	xx	) 00	8	8	x	œ	ø	4
	cm ⁻¹		e 18)		band	ν ₁	$\nu_1$	م ^ا	Γ. ¹ Λ	•	$\nu_1$	<u>-</u>	-		۳	•	$\nu_{1}$		$\nu_1$
1	2790	FIGURE 18	TABLE 2 (figure		ident.	(CH, CH,	QQH	OUH	(HDO	(CH₄	HDO	CH₄ HDO	CH,	$\operatorname{CH}_4^{1}$	(HDO	(CH₄	) HDO	(CH₄	HD0
	••		Ta	$\nu \; (\mathrm{cm}^{-1})$	(vac.)	2813-0	2811-0	2805-4 2805-4	0.000	7.0007	2801.6	5.797.9	2794.8	2793-7	0.0070	0.7617	0.087.6		2787.5
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					band	ν ₁	$\nu_1$			$\nu_1$	I	- <u>'</u>	-	٣	$2^{i}_{V_{2}}$	ν ¹	$\nu_1$		
_	2840				ident.	(HDO (CH,	(HDO	CH4 CH1	CH ¹	JHDO	(CH4	CH₄ HDO	CH,	(HDO	,HO,H	HÌDO	∫HDO	(CH4	•
				$\nu (\mathrm{cm}^{-1})$	(vac.)	2838.3	2835.0	2831.9	2830.2	9898.9	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	2822-3		C•7787	2819.8	2817-0	9815.0	0.0107	
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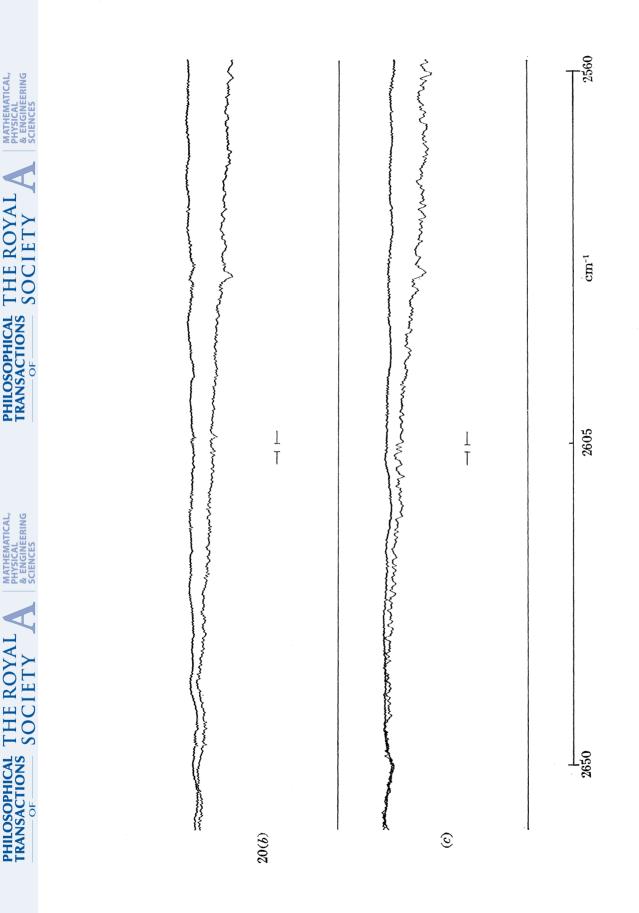


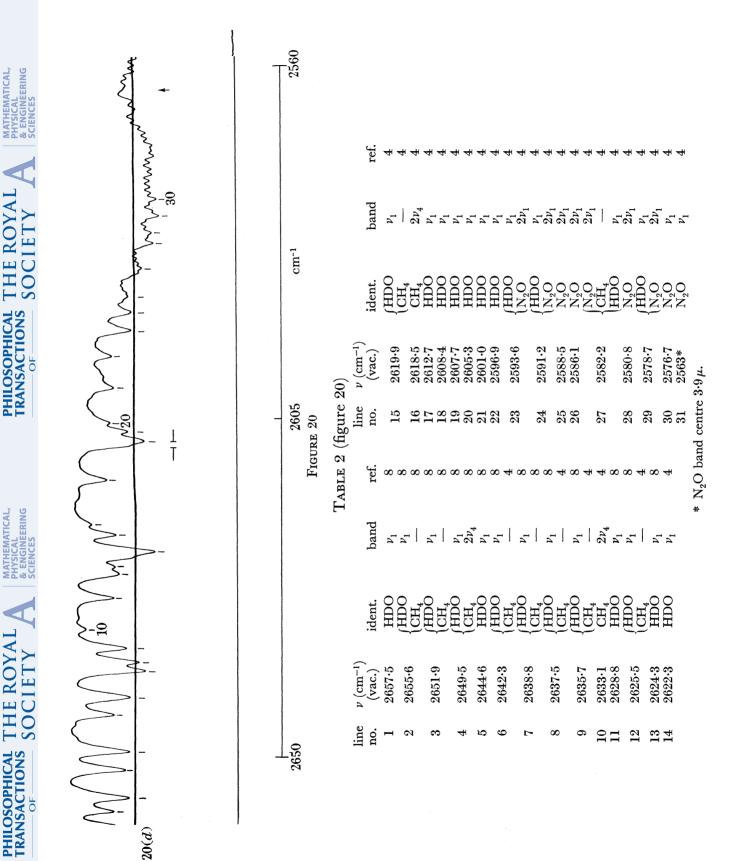


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		ref.	<b>x</b>	∞ ·	4 x	) X	ø	×	x	x	00	×	<b>∞</b> (	xx	o oc	×	x	8	x	x
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cm ⁻¹		ident.	$CH_4$	$\operatorname{CH}_{4}$	{CH4 {HD0	HDO	HDO	HDO	HDO	(HDO	{CH ₄	HDO	HDO O	(HDO	(HDO	CH,	, HDO	HDO	HDO	HDO
	(61	$\nu (\mathrm{cm}^{-1})$ (vac.)	2706.6	2704.8	2697.6	2695.5	$2693 \cdot 1$	2690.1	2680.9	0.0796	0.0107	2676.7	2675-4	2672.8		2669.9	2666.4	2663.3	2660.6	2659.6
2715 Figure 19	figure	line no.	20	21	22	23	<b>24</b>	25	26	5	11	28	29	30	10	31	32	33	34	35
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	$\mathbf{T}_{\mathbf{r}}$	band	$2 u_4$	$\nu_1$	۲ ^۱		$2 u_4$	$\nu_1$	•	-	μ,	, ¹	$2 \tilde{ u}_4$	۲ ۱	7.4	р. 1	, 'Y	р, г	$2\nu_{s}$	۳, ±
		ident.	$CH_4$	HDO	CH, CH,	$\operatorname{CH}_{4}^{1}$	$CH_4^{\dagger}$	HDO	CH4		HDO	OQH	(CH4		HDO	HDO	HDO	HDO	$CH_{A}$	НПÓ
		$\nu (\mathrm{cm}^{-1})$ (vac.)	2756.9	2753.8	2751.6 2750.0	2746.2	2741.8	2738-9	2737-4	2732.8	$2731 \cdot 1$	2729.9	2726-6	9.793.6	2722.2	2720-7	2718.9	2716.4	2712.0	2708.3
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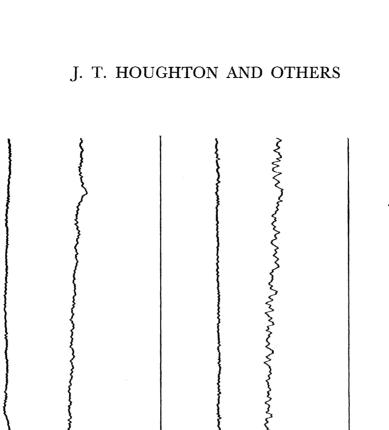
91

# ATLAS OF THE INFRA-RED SOLAR SPECTRUM





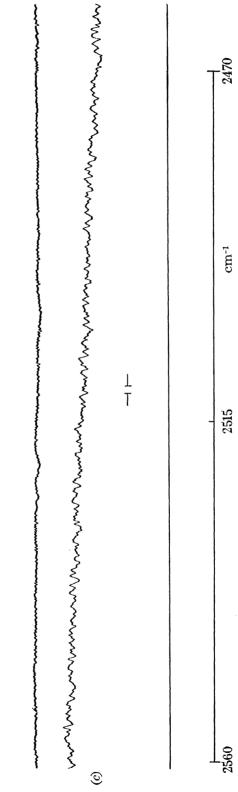


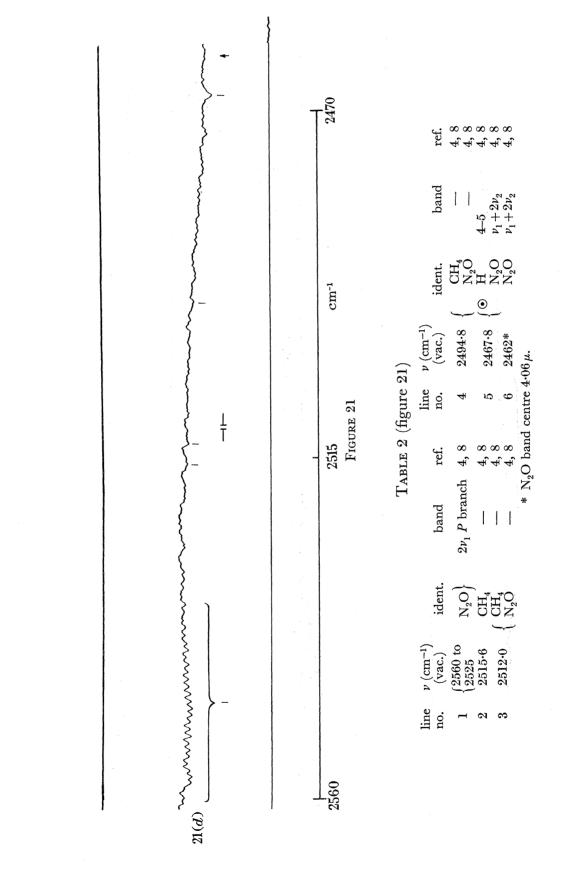


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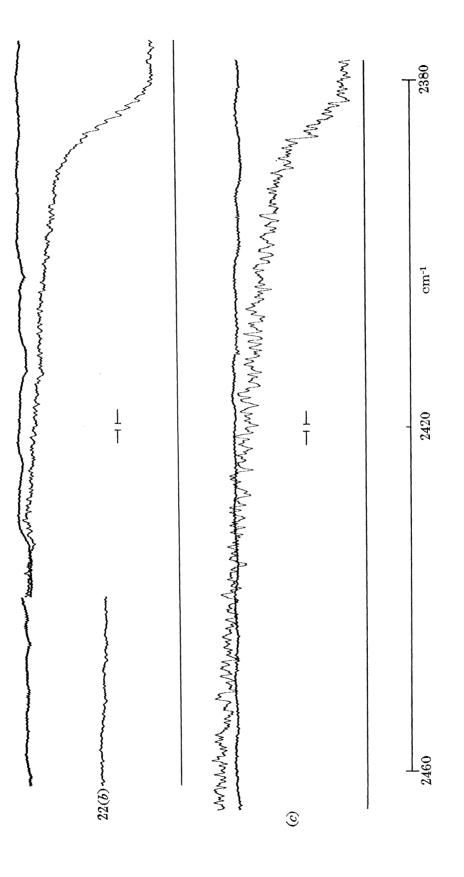


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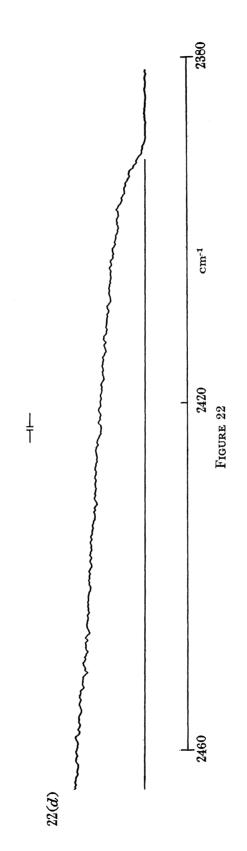
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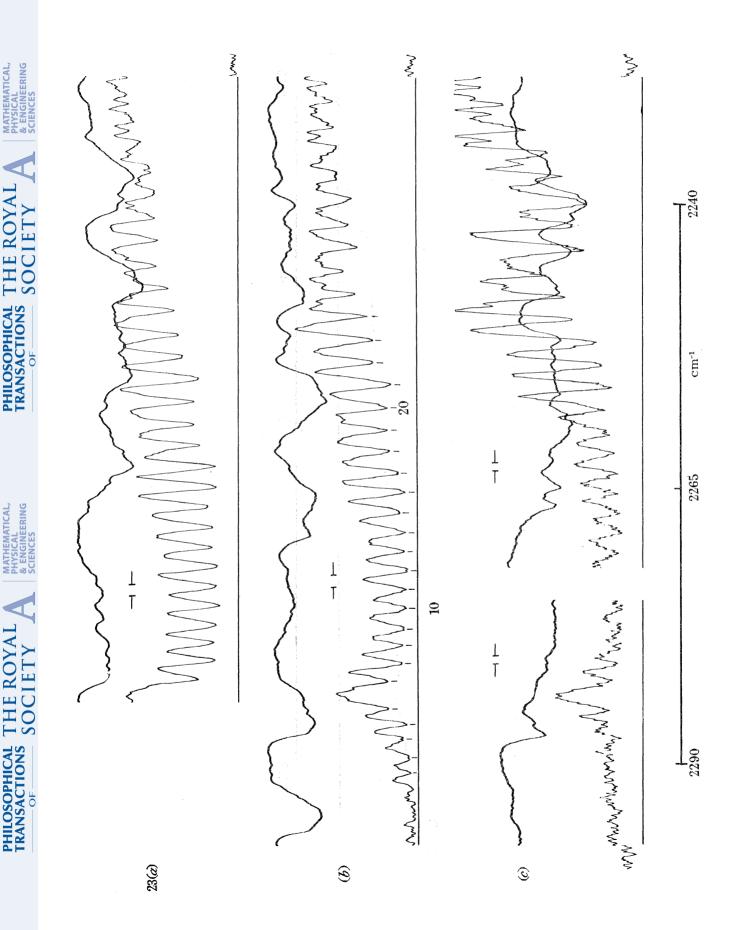
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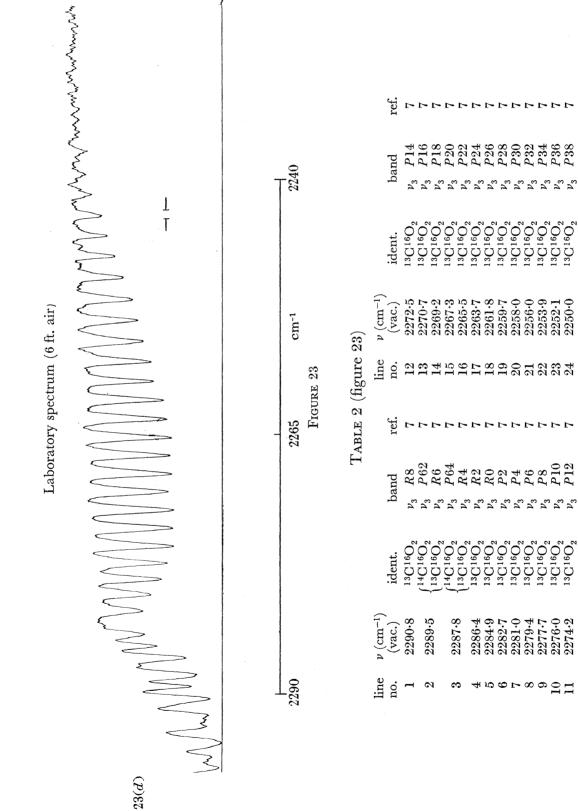
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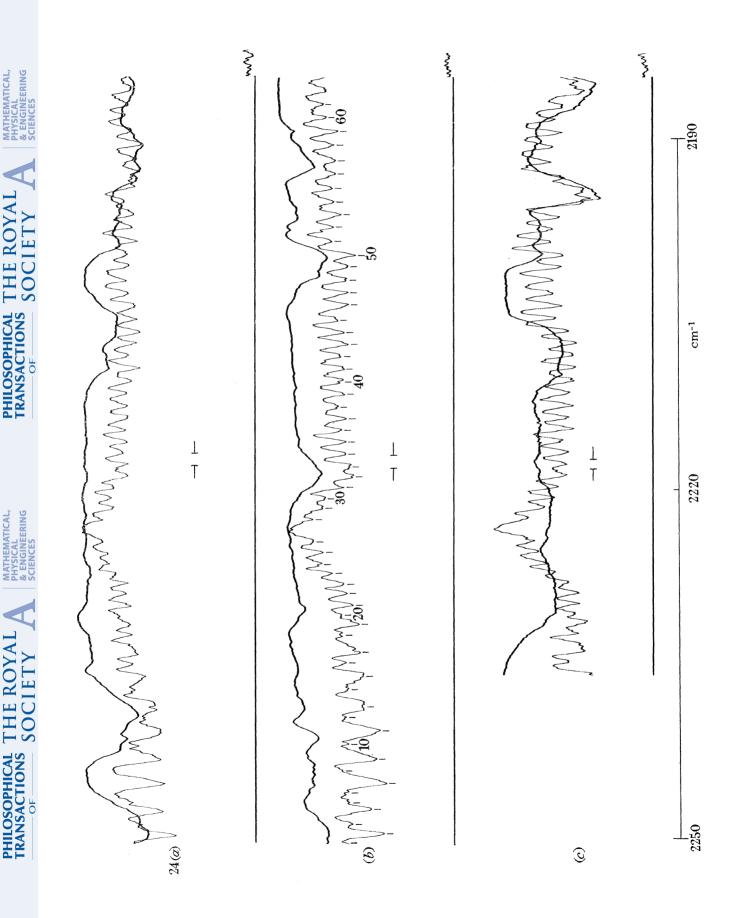
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#### ATLAS OF THE INFRA-RED SOLAR SPECTRUM





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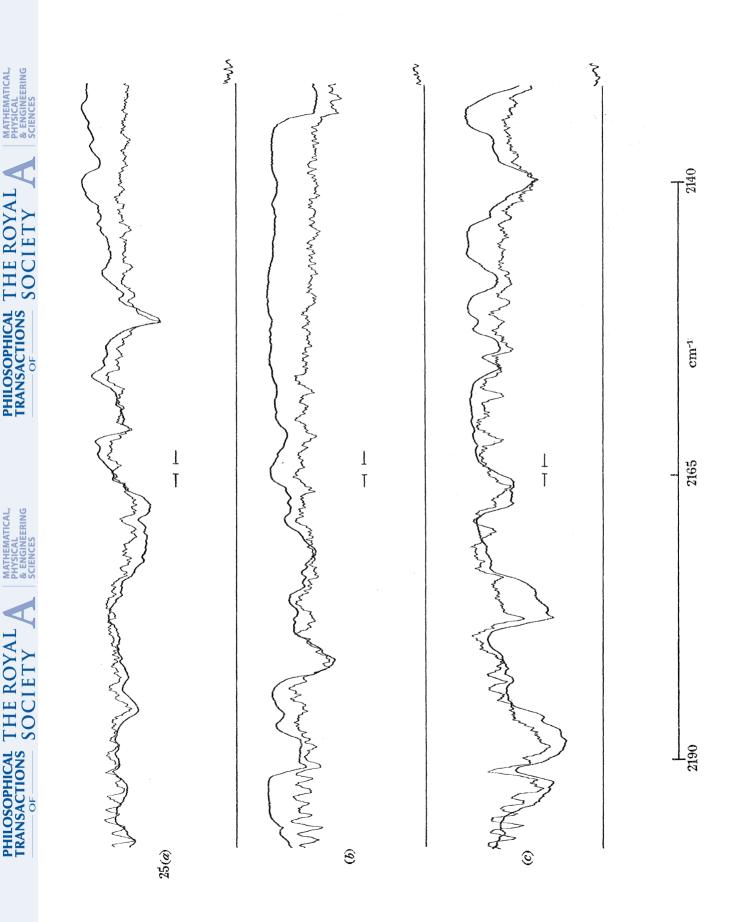
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cm⁻¹

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		ref.	4	44	4,	44	4	44	4	4	4	4	4 4	# <del>4</del>	4	4 4	╡╋	4,	44	4 4	4	44	4
		band	$p_3$ $P13$	$1-0 R19 \ v_3 P14$	$v_3 P15$	${\scriptstyle \nu_{3} \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \$	$\nu_{3} P17$	$\nu_{3} P18$	1-0 R17	$ u_3  P20 $	$\nu_3 P21$	$\nu_3 P22$	$\nu_{3} P23$	$v_{\circ} P24$	$v_3 P25$	$\nu_{3} P26$ 1_0 R15	$\nu_{3} P27$	$\nu_3 P28$	$\nu_{3} P29$ 1-0 R14	$\nu_3 P30$	1-0 $R13$	$v_3 P32$ $v_6 P33$	$v_{3} - P34$
		ident.	$\int N_2 O$	N,O	N ² 0	CO CO	N ₂ O	O ² N		N ₂ O	$N_2O$	$N_2^{-}O$	N ² O	O°N	$N_2^2O$	${\rm N}^{\rm N}_{\rm O}$	N ₂ O	N ₂ 0	CO CO			N ² O	$N_2^2 O$
		$\nu \left( \mathrm{cm}^{-1} \right) $ (vac.)	2212.3	2211.3	2210.4	2209.5	2208.6	2207-6	2206.7	2205-7	2204.7	2203.7	2202.7	2201.8	2200.8	2199.8	2198.7	2197-7	2196.7	2195.7 2194.6	2193.5	2192.4	2191.4
		line no.	40	41	42	43	44	45	46	47	48	49	50	51	52	53	54	55	56	57	20	209 90	61
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	e 24)	band		$v_{3} P54 v_{5} R11$		$v_3  R9 \\ v_8  R8  R8  R8  R8  R8  R8  R8  $			$v_3 R_5$	$\nu_3 R4$	$\nu_3$ R3	$\nu_3 R2$	$\nu_3 R1$	$v_3 P_1$	$\nu_3 P2$	$\nu_3 P3$	v3 P5	$\nu_3 P_6$	$v_3 P7$	$v_{3} P9$	$\nu_3 P10$	$v_3 P11$	1 1 20
FIGURE 24	TABLE 2 (figure 24)	ident.	$N_2O$	$ \underbrace{ \begin{bmatrix} 13C16O_2 \\ N_0O \end{bmatrix} } $		N2O	$\left\{\overline{13C16O_2}\right\}$	N20 N	N.O	N,O	$N_2O$	$N_2O$	O ^z z		$N_2^2O$	N ² O	N ² O	N ₂ O		[N ² 0 00	N ₂ O	N2C	) 7
	TA	$   \nu \left( { m cm}^{-1} \right)   \left( { m vac.} \right) $	2234.0	2233.3		2231.8	2230-9	2230.2	2228.6	2227.8	2227.1	2226.2	2225.5	2222-9	$222 \cdot 0$	2221-2 9990.4	2219.5	2218.6	2217-7	2215.9	2215.1	$2214 \cdot 1$ $2213 \cdot 2$	
		line no.	16	17	( ,	<b>X</b>	Гĥ	20 91	55	23	24	25	26 26	58- 78- 78-	29	30 21	32	33	48 47 72	36	37	88 80 80	0
		ref.	4	44	4,	44	4	44	4	4	4	4	4.	<del>1</del> 4	4	4~	H 4	4.	44	44	4	44	4
		band	$\nu_{3} R34$	$\nu_{3}^{\nu_{3}} P40 \ \mu_{5} R33$	$\nu_3 R32$	$v_3 K31 v_5 P42$	$\nu_{3}^{2} R30$	$\nu_{3} R29$	$v_3 R27$				$v_{3} R24$							$v_3 R17$			
		ident.	N ₂ O	$\begin{cases} {}^{13}C^{16}O_2\\ N_0O \end{cases}$	$(N_2^{2}O)$	N2U [13C16O,	$\{N_2O_2$	N2O	N ² 0	( ¹³ Č ¹⁶ O,	$\left\{ N_{2}O^{2}\right\}$	$(N_2^{-}O$	N2O		$(N_2^2O)$		N ₂ O ²	$(N_2O$	N2O [13C16O_		N2O		$\left( \frac{1}{N_{2}^{2}O} \right)$
		$\nu \left( \mathrm{cm}^{-1} \right) $ (vac.)	2248.7	2247.7		2246-9	2245-7	0.346.0	2244·4		2243.7		2242.4	2241.6	1	2240.5	2239.5		2238-4	2237-0	2236.3	2234.9	
		line no.	I	69	d	no A	4	ĸ	. v		2		œ	6		10	11	( ,	12	13	14	15	)

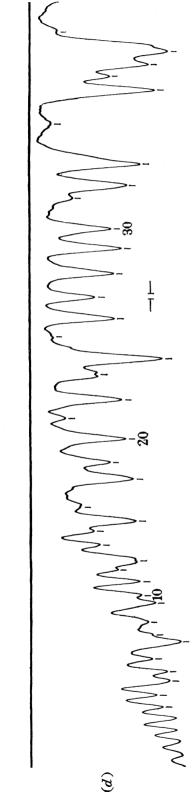


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2140	
cm ⁻¹	
2165	FIGURE 25
2190	

		-																	
		band	1-0 R3	$ u_2 $	1-0 R2	ν,	1-0 $R1$		I-0 R0	ν,	l=0 R	I-0 PI	V,9	ν,	1-0 P2	1-0 R			
		ident.	CO	$\rm H_2O$	80	Н,О	CÕ		SO	Н,О	13Č16O	80	H,O	$H_{0}$	Q	13C16O			
	$\nu (\mathrm{cm}^{-1})$	(vac.)	2158.4	2156.7	2154.7	2152.6	2151.0	2148.5	2147-4	2145.6	2141.3	2139.5	2138.3	2137.3	2135.8	2134.4			
	line	no.	26	27	28	29	30	31	32	33	34	35	36	37	38	39			
		ref.	4	4	4	4	4	4	4	4	4	4	4	4	4	4	4	4	
re 25)		band	$\nu_3 P46$	$\nu_3 P47$	1-0 R8	$\nu_{s}$ P48	$\nu_{s}$ P49	1-0 R7	$ u_3  P50 $	$\nu_{3}$ P51	ν,	$\nu_3 P52$	1-0 R6	ν,	1-0 R5	$ \nu_9 $	1-0 R4	$ u_3 $	•
TABLE 2 (figure		ident.	N,O	N,O	<u>c</u> o	(N,O	N,O	[CÔ	0°N	N,O	ÌΗ,O	(N,O	ÇÕ	H,O	CŐ	Н,О	CÕ	$N_2O$	1
$T_A$	$\nu (\mathrm{cm}^{-1})$	(vac.)	2178.3	2177-1	0 0110	C.0/17	2174.7		2172.9		1 1210	<b>7.1/17</b>	2169.3	2167.5	2165.7	2163.5	2161.9	2159.6	
	line	no.	14	15	υL	01	17		18		OL.	гя	20	21	22	23	<b>24</b>	25	
		ref.	4	4	4	4	4	4	4	4	4	4	4	4	4	4	4	4	4
		band	$\nu_{3}$ P33	$\nu_3 P34$	1-0 R12	$\nu_{*} P35$	$\nu_{s}$ P36	$v_{3}$ P37	1-0 <i>R</i> 11	$\nu_{*} P38$	$\nu_{s}$ P39	$\nu_{3}$ P40	1-0 R 10	$\nu_{s} P41$	$\nu_3 P42$	$\nu_3 P43$	$\nu_3 P44$	1-0 R9	$\nu_3 P45$
		ident.	N,O	N,O	Ō	(N _o O	N,O	N,O	ſĊŐ	O°N)	N,O	N,O	,cõ	(N,O	N,O	N,O	N ₂ O	Ō	$(N_2O$
	$\nu \; (\mathrm{cm}^{-1})$	(vac.)	2192.6	2191.6	1 0016	2190.4	2189.6	2188.4	0 1010	0.1812	2186.1	2185.1	7 6016	0.2012	2182.8	2181.6	2180.4	9170.0	0.6117
	line	no.	Г	61	c	o	4	ũ	c	0	7	8	c	ת גי	10	11	12	61	01 •
														v	)L	25	14.		<b>3.</b>

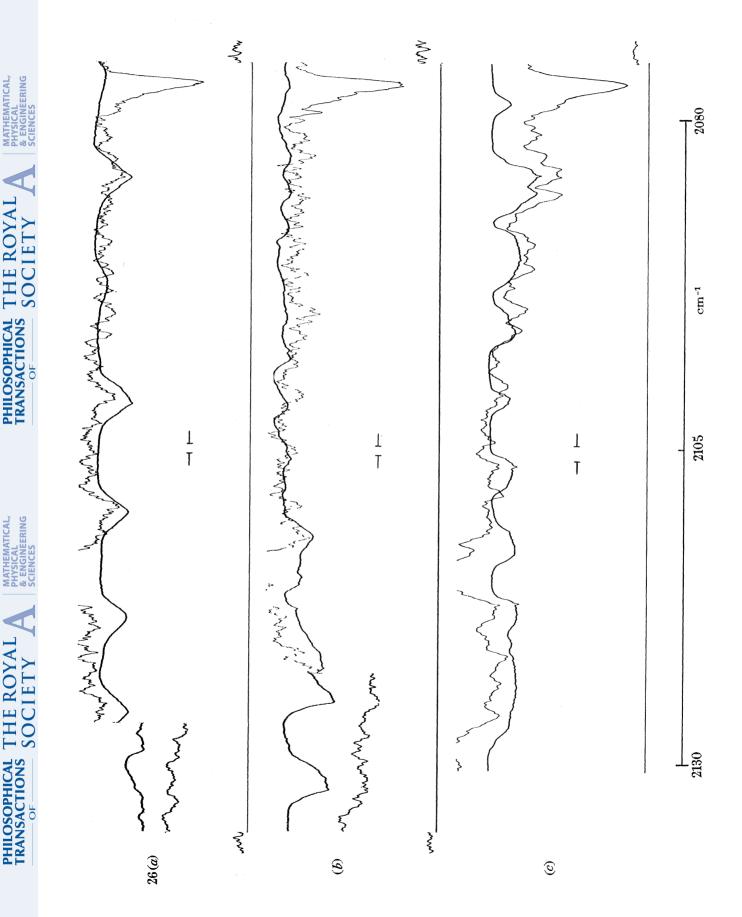
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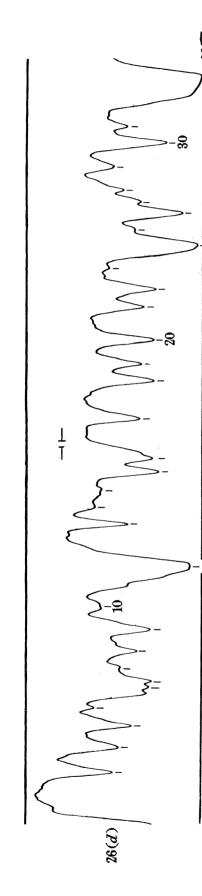
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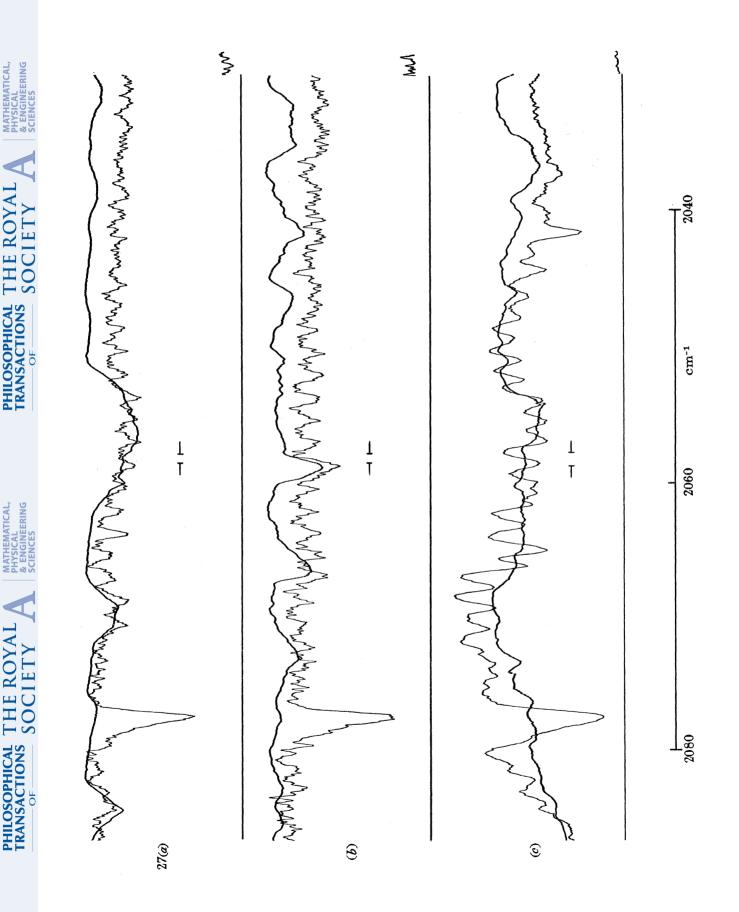




0					7	COIT		cm ⁻¹				2080
			ı		FIGUR	FIGURE 26						
				Tai	LABLE 2 (figure	figure <b>;</b>	26)					
line	$\nu (\mathrm{cm}^{-1})$	:			,	line	$\nu (\mathrm{cm}^{-1})$				,	
no.	(vac.)	ident.	banc		ref.	no.	(vac.)	ident.	band		ref.	
I	2131.6	CO	1–0		4	17	2103.2	CO	1-0	P10	4	
2	2129.6	c0,	$2\nu_1 - \nu_2$	ò	4	18	2100.5	H,O	ν,		4	
ი	2127-7	00	1– <u>0</u>		4	19	2099.1	CÕ	$1^{-0}_{-0}$	P11	4	
4	2126.1	co,	$2\nu_1 - \nu_2$		4	20	2097.3	O,H	ν,		4	
		$\int 1^{3}C^{16}O$	$1-\hat{0}$		4	21	2094.9	CŐ	$1_{-0}$	P12	4	
ñ	2124.2	CO,	$2\nu_1 - \nu_2$		4	22	2093.5	CO,	$\nu_1 + 2\nu_5^2 -$	- V,	4	
		(H,Õ	ν, -		4	23	2091.7	co,	$\nu_1 + \nu_2$	$R$ $\mathbf{\tilde{l}8}$	4	
9	2123.6	CÕ	$1^{-0}_{-0}$	P5	4			ĺΗ,Ô	V, 2		4	
Г	0100.7	$ ho H_2O$	$\nu_3 - \nu_2$		4	24	2090.6	{CÕ2	$\nu_1 + \nu_2$	R16	4	
-	1 1 1 1 1	lCO3	$2\nu_1 - \nu_2$		4				1-0 -	P13	4	
x	2121.3	$(\mathrm{H_2O}$	$\nu_2$		4	25	2088.5	CO2	$\nu_1 + \nu_2 R 14$	R14	4	
Ď		$(CO_2)$	$2\nu_1 - \nu_2$		4	9.6	2087.1	$\int CO_2$	$\nu_1 + \nu_2$	R12	4	
<b>0</b>	2119.7	00	1-0	P6	4	01	T. 1007	(H,O	$\nu_9$		4	
10	2118.0	်			4	27	2086.3	CÕ	$1^{-0}_{-0}$	P14	4	
ΙI	2115.6	Õ	1-0	P7	4	28	2085.4	CO,	$\nu_1 + \nu_2$	R10	4	
12	2111.6	SO	1-0	P8	4	29	2083.8	co,	$\nu_1 + \nu_2$	R8	4	
13	2110.2	c0,	$2\nu_1 - \nu_2$		4			[CO ²	1-0	P15	4	
14	2108.7	co,	$2\nu_{1}^{2}-\nu_{3}^{2}$		4	30	2081.9	CO,	$\nu_1 + \nu_2$	R6	4	
15	2107.4	CO	<b>1</b> -0	P9	4			, Η Ô	V., 2		4	
0 -		( ++						4	4			

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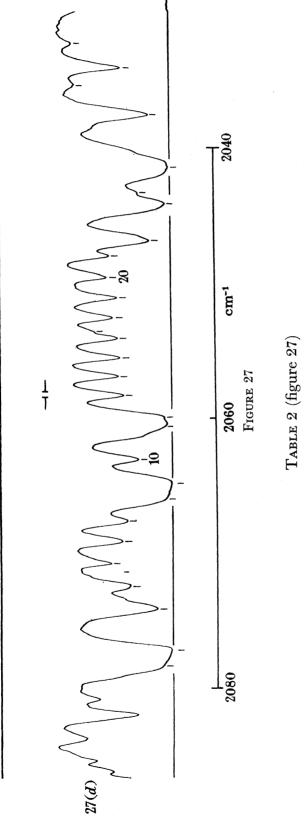




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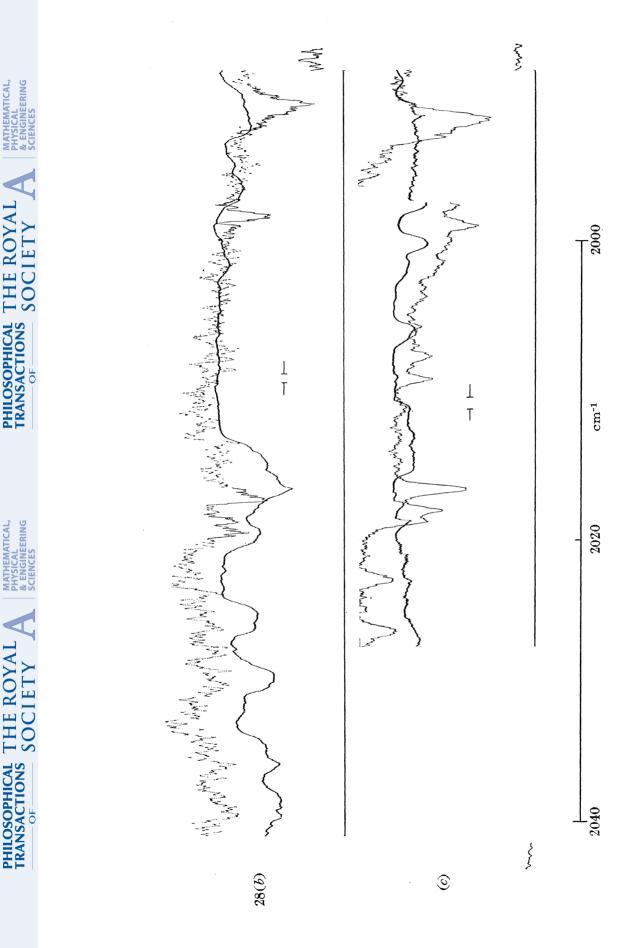


		X																
		band	$\nu_1 + \nu_2  P34$	$\nu_1 + \nu_2  P36$	$\nu_1 + \nu_2  P38$	$\nu_1 + \nu_2  P40$	V2 5.0	$\nu_1 + \nu_2 P42$	$\nu_1 + \nu_2  P44$	$\nu_1 + \nu_2  P46$	$\nu_2$	$\nu_2$	$\nu_1 + \nu_2 P54$	V2	$\nu_1 + \nu_2 - P56$	$v_1 + v_2 P58$		
										CO2								
	$\nu (\mathrm{cm}^{-1})$	(vac.)	2050.5	2049.0	2047.5	2046.0	2044-4		2042.9	2041-3		2037.4	2035.4	2034-0		2032.3		
	line	no.	19	20	21	22	93	2	24	25	2	26	27	2.8	2	29		
		ref.	4	4	4	4	4	4	4	4	4	4	4	4	4	4		
		band	ν,							1-0 P20							1	
TABLE 2 (IISUIC 21)		ident.	(H,O)	{CÔ,	CO	, CO,	$\operatorname{CO}_{2}^{2}$	(CO,	{H,Õ	(cô	CO2	CO.	CO2	$CO_2$	H,18O	çõ,	1	
T	$\nu \; (\mathrm{cm}^{-1})$	(vac.)		2064.5		2062.7	2061.3		2059.8		2058.2	2056.6	2055.2	2053.6	2052.9	$2052 \cdot 1$		
	line	no.		6		10	11		12		13	14	15	16	17	18		
										4								
		band	1-0 $P16$	$\nu, +\nu, R0$	$v_{1} + v_{2} = 0$	$v_{0} - v_{0}$	l_0 [^] P17	ν,	$v_{1}+v_{2}$ P4	$\nu_1 + 2\nu_2^2 - \nu_2$	$\nu_1 + \nu_2 P \tilde{6}$	$\nu_1 + \nu_2 P 8$	1-0 ² P18	$\nu_1 + \nu_2 P I 0$	$\nu_1 + \nu_2 P I 2$	$v_1 + v_2 P I 4$	$\nu_2$	
		ident.	COD			(H, O	ĺcÔ	H _o O	CO.		Ć,	CO,	CO	(co,	ÇO [°]	CO,	$\{H_2 \hat{O}\}$	I
	$\nu  ({\rm cm}^{-1})$	(vac.)		2077-3		2076-9	2073-2				2072.2	2070.6		2068-9	2067-5		2065-9	
	line	no.		-		51	က				4	20		9	7		x	

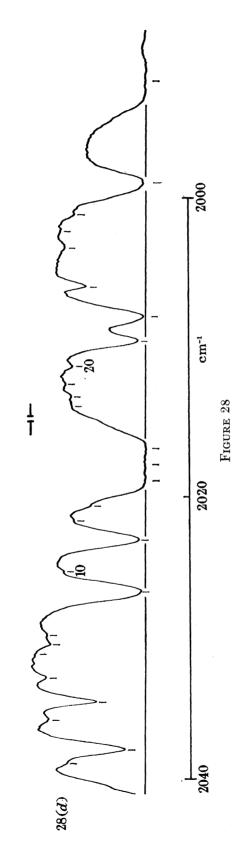
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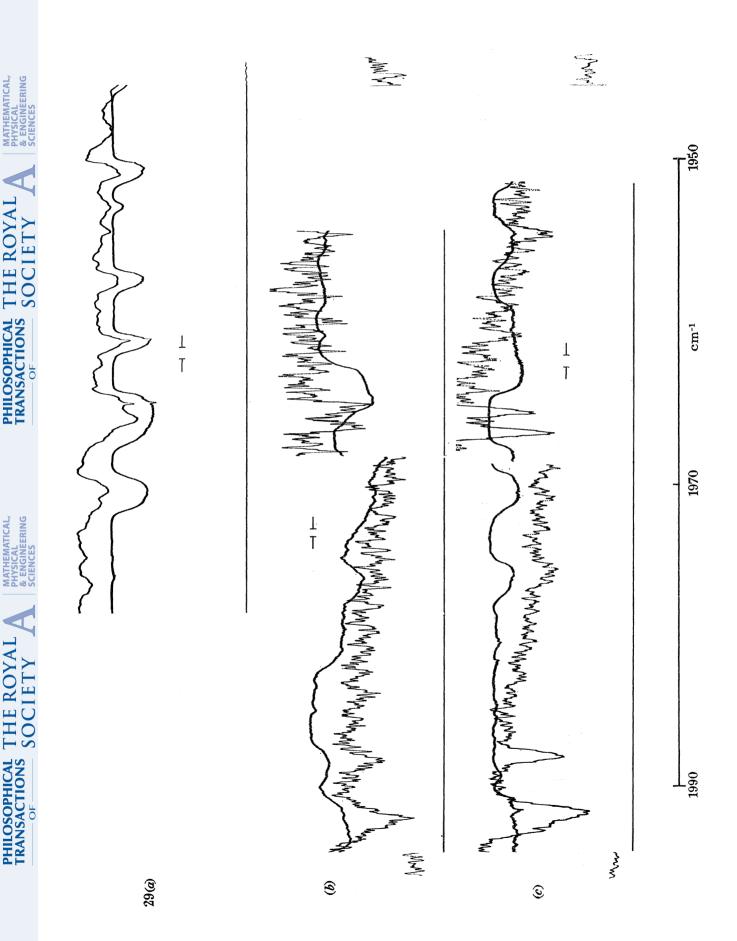
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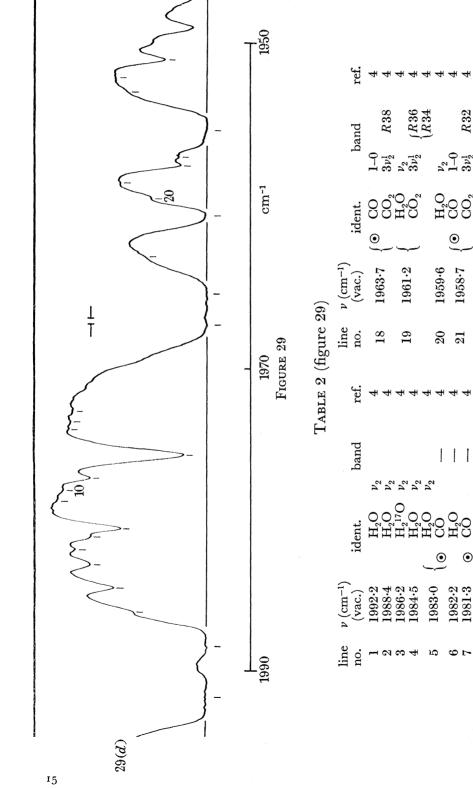




	c	ret.	4,	44	4	4	4	4	4	4	4	4	4	4	4	4	4	4
		ident. band	$H_2O$ $\nu_2$	$H_2O  \nu_2$ 13 $C16O  \nu \perp \nu$	$CO \sim 2 \cdot 1 + 2$	$^{13}\text{C}^{16}\text{O}_2 \ \nu_1 + \nu_2$	$^{13}\text{C}^{16}\text{O}_2 \ \nu_1 + \nu_2$	$H_2O = \nu_3^2 - \nu_2^2$	<u>Q</u>	$ m H_2O$ $ u_2$	$\mathrm{H}_2^{-}\mathrm{O}$ $\nu_2^{-}$	$\mathrm{H_2^{18O}}$ ? $\nu_2$	$H_2O$ $\nu_1 - \nu_2$	CO 1-0	CO	$H_2O$ —	$\mathrm{H}_{2}^{-}\mathrm{O}$ $ u_{2}$	$\mathrm{H_2O}$ $\nu_2$
				Ļ	$\sim$		ت	~	•				_	$\tilde{\mathbf{O}}$	۲			
28)	$\nu (\mathrm{cm}^{-1})$	(vac.)	$2018 \cdot 3$	2016-8	$2015 \cdot 1$	2013.5	0100	<b>7.</b> 7107	2011.4	2009.3	2007.6	2005.6	0003.9	10007	2002.1	2000.9	1998-9	1992.2
figure 2	line	no.	15	16	17	18	01	ля	20	21	22	23	16	H	25	26	27	28
LABLE 2 (figure		ref.	4	4 -	₽₹	4	4	4	4	4	4	4	4	4	4	4	4	4
$\mathbf{T}_{\mathbf{A}}$	,	q	P50	DEG	P54		P56	P58	P60		P62							
		ban	$\nu_1 + \nu_2 P50$	ν ₂ "' ' "	$v_1 + v_2 + v_3$	7°,	$\nu_1 + \nu_2$	$\nu_1 + \nu_2$	$\nu_1 + \nu_2$	ν, ²	$\nu_{1} + \nu_{2}$	$\nu_{1} + \nu_{2}$	$\nu_2^{-1}$	$1_{-0}$	$\nu_{9}$	$\nu_{1} + \nu_{2}$	$\nu_{1} + \nu_{2}$	$\nu_2^{-1}$
	:	ident.	$\overline{\mathrm{CO}}_2$	H ² O		H"Ó	CŐ	CO,	co,	$H_{,18}$ O	CÔ,	13C ¹⁶ O ₂	$H_{2}O$	13C16O	H,O	13Č16O,	13C16O,	$H_2O$
	$\nu (\mathrm{cm}^{-1})$	(vac.)	2038.5	$2037.4$ {	2035-4	J 077600	0.4007	2032·3	2030.9	2030.2	J 6.0006	e.6707	2026.6	2025.0	2023.0	2021.5	2019.7	2019.0
	line	no.	Ţ	61	co	Ţ	<del>1</del>	5	9	2	0	0	6	10	11	12	13	14





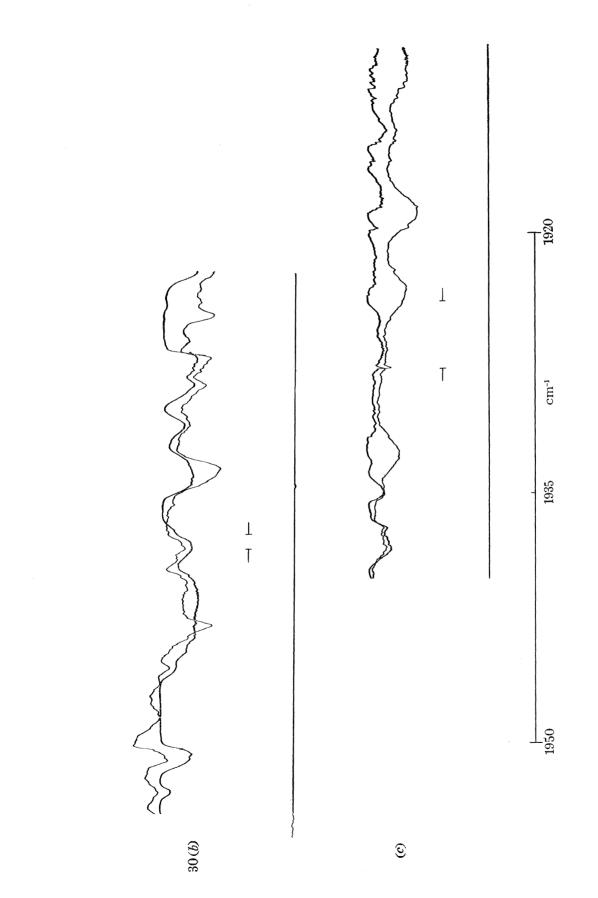


		ref.	44	44	44	44	4	4 4	44	44	4	44	
		and	R38	${{\nu_2}\atop{3{\nu_2}}}^{\prime_2} \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \$	(K34	R32		R30	R28	[R26	R24	R22	
		dent.	000	$H_2 \tilde{O}_2$	О"Н	ပ္ပံပိ	$H_2 \tilde{O}$	CO, EO	Н <u>,</u> О СО,	, OO	CO ₂	H ₂ O	N
		.4		$\smile$		<u> </u>	, ,	<u> </u>	$\overline{}$	•		<u> </u>	,
igure 29)	_			1961.2									
	line	no.	18	19	20	21	22	23	24	25	$\overline{26}$	27	
ABLE 2 (figure		ref.	44	44	44	44	4	44	44	44	4	44	4
T		band	$V_2^{V_2}$		ν ₂		$\nu_{3} - \nu_{2}$	$^{ u_2}_{1-0}$	ν, 	ν2 	1-0	ľ.	$\nu_2^2$
		ident.	${ m H_2^{00}O}$	$H_2^{17}O$	⊙ COO	о СО СО	$\widetilde{H_{20}}$	⊙ CO CO	⊙ CO H,¹8O	• CO CO	000	⊙ H°O	$H_2^2O$
	$\nu \; ({ m cm}^{-1})$			1986.2 1984.5									

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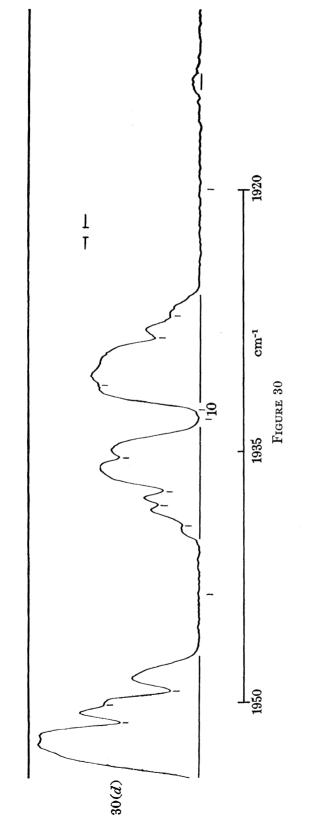
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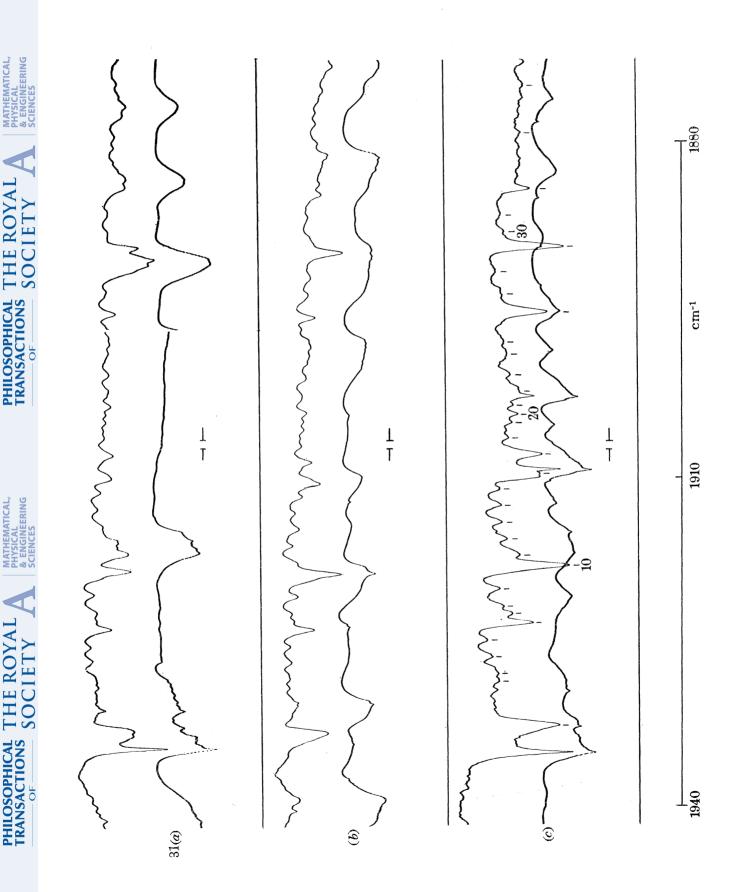
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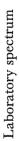
	ref.	44	4	4 4	4	4~	₩ <del>4</del> 1	
	band	$v_2^{\nu_2}$	v2 &		$\frac{v_2}{3v_2^1} P6$	ν2 3.1 DS	Ur2 10 Vs	4
	ident.	H20 CO2	$H_2^{18}O$	CO H_O	$\{ CO_2^{\circ} \}$	$\left\{ \begin{array}{c} H_{2}O\\ OO\\ OO\end{array} \right\}$	H,O	1
30)	$     \nu  (\mathrm{cm}^{-1}) $ (vac.)	1933-2	1932-1	1930-6	1927-8	1926.7	1920	
(figure	line no.	6	10	11	12	13	14	
ABLE 2 (	ref.	44	1 <del>4</del> 1 -	4 4	4	4	┞╼╢	4
[1	band					v2 3.,1 P.6		1
	ident.	$H_{CO}^{H_2O}$	H ₂ O	H ² O	$H_2O$	H ₂ O	H,O	${ m H_2^{ m O}}$
	$\nu (\mathrm{cm}^{-1})$ (vac.)	1951-1	1950-1	1949-3	1943	1939-2	1937-2	1935.4
	line no.	I	67	က	4	ນດ	5	œ

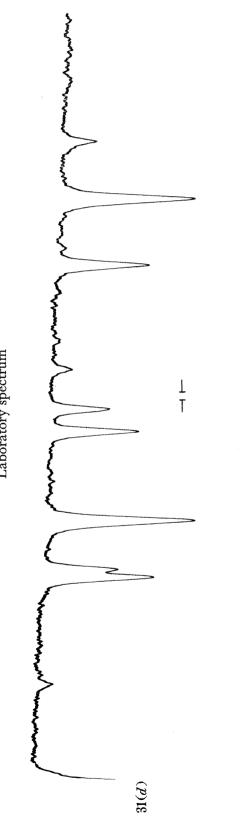
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		band		$v_2 y_1^2 P48$									$\nu_2^{-}$	$\nu_2$	$\nu_2^-$		
		ident.	$CO_2$	Ч20 СО О	CO.	CO,	CO2	H ₂ O	CO2	CO2	$CO_2$		${ m H_2O}$	$H_{2}^{-}O$	$\rm H_2^{-}O$		
		$\nu (\mathrm{cm}^{-1})$ (vac.)	1897.9	1895.3		1893.2	1891.7		1889.6		1887-4	1886.0	1884.6	1879.4	1876-7		
		line no.	25	2.6	i	27	28		29		30	31	32	33	34		
		ref.	4	44	·	-	I	Ţ	4	4	Г	4	4	T	4	4	
	1)	band	P24	$\frac{3\nu_2^{ m I}}{3\nu_1}  P26$	1	P30		P32	P34	P36		P38	P40		P42	P44	d.
31	ure 3		$3  u_2^1$	$3\nu_1^1$	V.2	$3\nu_{1}^{2}$	r, 2	$3\tilde{\nu}_{2}^{1}$	$3\nu_2^{ ext{I}}$	$3\nu_{2}^{1}$	ν.,	$3\tilde{\nu}_{1}^{1}$	$3\nu_5^1$	ν. ⁻	$3 \overline{\nu}_2^1$	$3\nu_2^1$	d value
FIGURE 31	TABLE 2 (figure 31)	ident.	co,	'0°C	H ₀	CÔ,	H,Ó	CŐ	CO,	CO,	Н"Õ	CŐ	CO,	Н,Õ	CÔ2	$CO_2$	Interpolated value
	TAB	$\nu \left( \mathrm{cm}^{-1} \right) \left( \mathrm{vac.} \right)$	1914-1	1912-6 1911-0	J	10161		T.ONAT	1906-5	1905.0	1904.5	1903.6	) 10001	1.706T	1900.6	1899.2	* II
		line no.	13	14 14		16		T (	18	<b>19</b>	20	21	00	77	23	24	
		ref.	-	4 4	4	4	4	F	H	I	4	4	H	F	I	4	
		hand		$3 ilde{ u}_2^1 \ Q$													
		ident.	Н,О	çç Çç	_	CO.	CO,	H,Ő	CÔ,	H,Ő	CÔ,	co,	H,Ô	CÔ,	co,	$\mathrm{CO}_2^{-}$	
		$\nu \left( \mathrm{cm}^{-1} \right)$	J 1 0001	~~		1926.2	1924.7	- J - 1 - 1 - 1 - 1 - 1 - 1 - 1 - 1 - 1	1923-1	1922-4	1921.7	1920-0	1 0101	1.8161	1917.2*	1915-5	
		line	-		۹ <b>۲</b>	94	1 10		9	7	8	6	•	10	11	12	

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1880

cm⁻¹

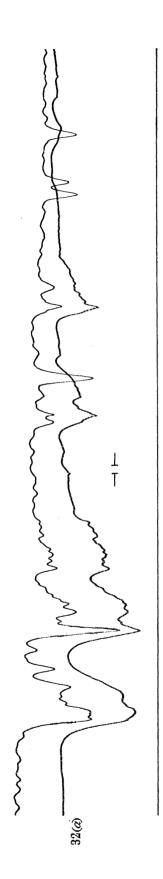
1910

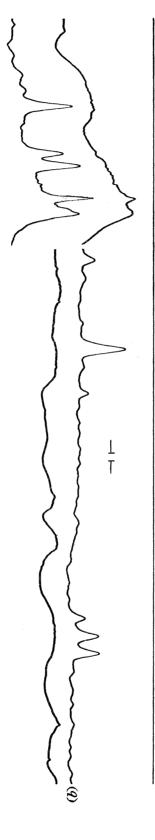
1940

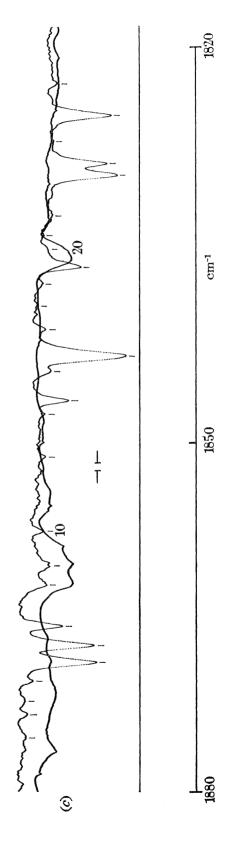
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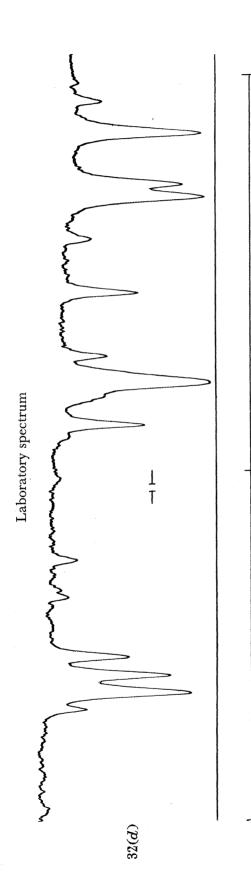












		د	ret	-	Τ	Τ	Π	T	I	Η	4	<b>i</b>	T	Η	4	Π	Τ	
FIGURE 32		-	band	$ u_2 $	$\nu_2$	<i>2</i> ،	$2\overline{\nu}_2 - \nu_2$	$\nu_2^{-}$	$\nu_2$	$\nu_2$	$\nu_2$	$\nu_2$	$\nu_2$	$\nu_2$	,	$ u_2 $	$\nu_2$	I
			ident.	${ m H_2O}$	$H_{2}^{-}O$	$H_{\overline{0}}$	$H_{0}$	$H_{2}O$	$\int H_2 \overline{O}$	( H ₂ ¹⁸ O	$H_{2}^{18}O$	$\rm H_2O$	$H_2O$	$H_{\overline{0}}$	00 ©	$H_2O$	$H_{2}O$	ı
	32)	$\nu (\mathrm{cm}^{-1})$	(vac.)	1844.2	1842.3	1840.5	1839-3	1837.3	1026.1	T.OCOT	1834.8	1833.4	1830.3	1829.4	1827-1	1825-3	1822.8	,
	TABLE 2 (figure :	line	no.	15	16	17	18	19	00	07	21	22	23	24	25	26	27	
		c	rei.	I	4	4	Ī	I	I	I	I	l	I	I	I	I	4	<b>–</b>
		-	band	ν,	,	•	ν,	ν°	r,	<i>ب</i> ،	r 64	<i>2</i> , <i>6</i> , <i>7</i> , <i>6</i>	ν <u>,</u>	<i>ب</i> ،	ν,	72 T		$\nu_2$
		:	ident.	H ₀ O		00		H,O	H,O	H,O	H,O	$H_{0}$	$H_{2}^{-}O$	H _. O	H,O	${ m H}_2^{ m O}$	00000	${ m H_2O}$
		$\nu (\text{cm}^{-1})$	(vac.)	1876.7	1873.5	1872.5	1870.8	1869.3	1867.9	1866.4	1864.1	1861.5	1858.5	1852.4	1848.8	J 8.7.81	0.1±0T	1845.4
		line	no.	I	61	က	4	20	9	1	x	6	10	11	12	12	P1	14

1820

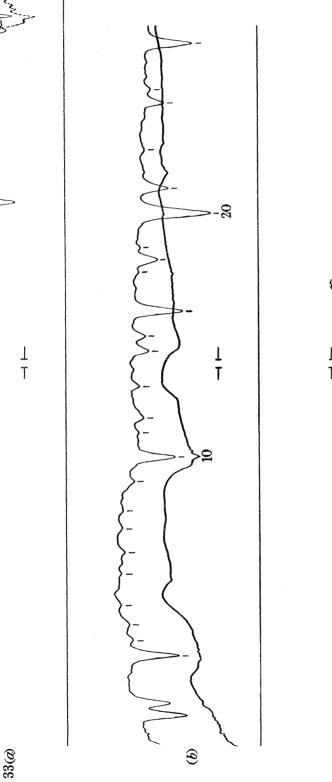
cm⁻¹

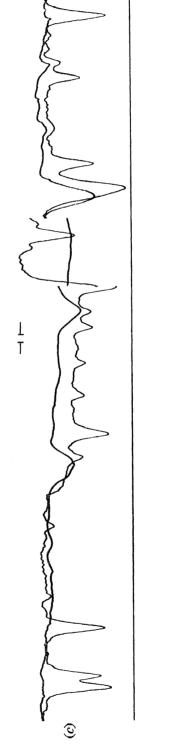
1850

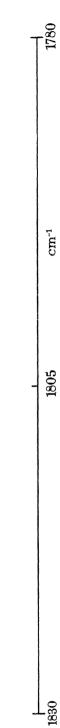
1880

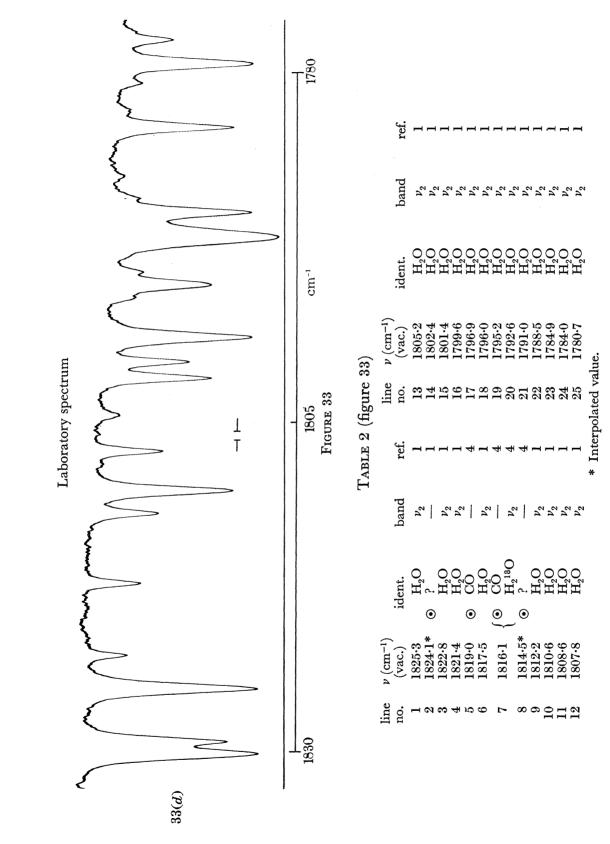
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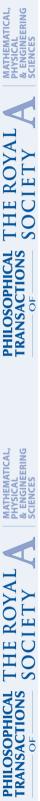


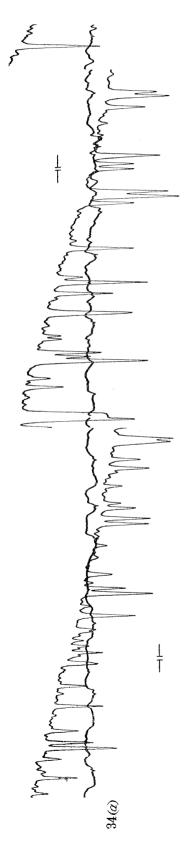




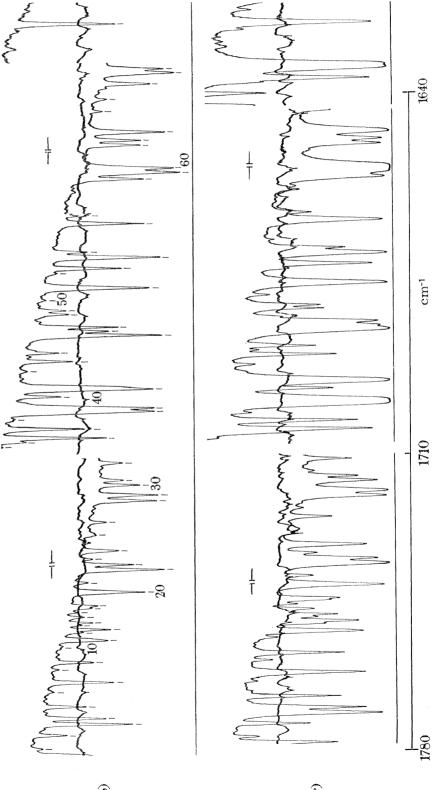








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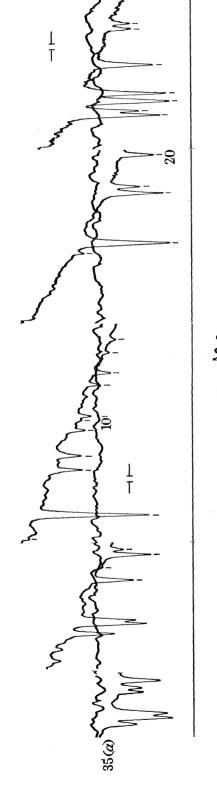
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PHI TRA	Laboratory spectrum	FIGURE 34	1       1.1       )       ident. $1^{-1}$ 3*       H2       (figure 34)         3*       H2O       -         9       H2O       -         8       H2O       -         8       H2O       -         8       H2O       -         8       H2O       -         9       H2O       -         6       H2O       -         9       H2O       -         1       H2O       -         9       H2O       -         9       H2O       -         1       H2O       -         9       H2O       -         1
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PHILOSOPHICAL TRANSACTIONS	34(d)	1780	$^{ u}$ (cm ⁻¹ ) (vac.) 1779-1 1775-6 1771-4 1771-4 1768-2 1768-2 1761-9 1761-9 1761-9 1761-9 1761-9 1765-3 1761-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1748-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-6 1778-7 1778-6 1778-6 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-7 1778-
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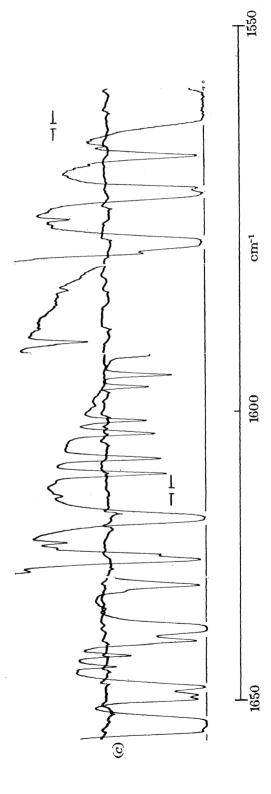
16-2



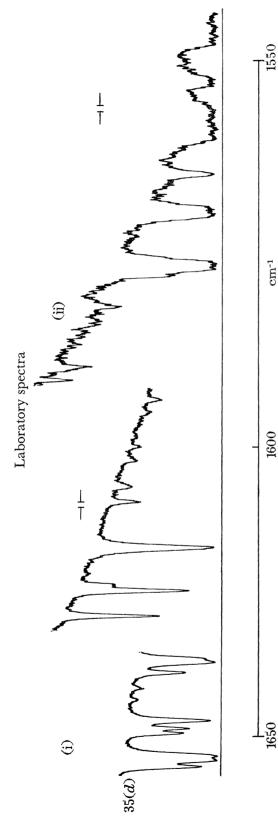
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Able 2 (figure 35)	band	V2 V3	7 2 2 7 2	ζζζ	2°2;	, ⁷ 2	72 29	$\nu_2$	$\nu_2$	
	ident.	H ₂ O H,O	H ₂ 0 H,0	H20 H20	H ₂ O	H ² O	Н20 Н,0	$H_{O}^{2}O$	H2O —	
	$\nu (\mathrm{cm}^{-1})$ (vac.)	1589.0 1577.6*	1576.2 $1569.8$	1569-0 1565-1	1560-3 1550-6	1558-9 1558-9	1554.4	1550.3	1545.3 1545.3	٥
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[-]	band	$\nu_2$	۲2 ۲3	*   ×	2 ² :	7 7 7 7	P2	$\nu_2$	7 2 2 2	*
	ident.	$\Theta$ H ₂ O	Н ₂ 0 Н,0	H ₂ 0 H ₂ 0	H2O	H20 H20	O _c H	$H_{O}^{2}O$	$H_2^{\rm H}O$	
	$\nu  (\mathrm{cm}^{-1}) $ (vac.)	1628.3 1626.2*	$1624 \cdot 2$ $1623 \cdot 3$	1621-5* 1617-4	1610-2	1603.7 1603.7	1601.6	1596-7	1592·3	
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1550

cm⁻¹

FIGURE 35